



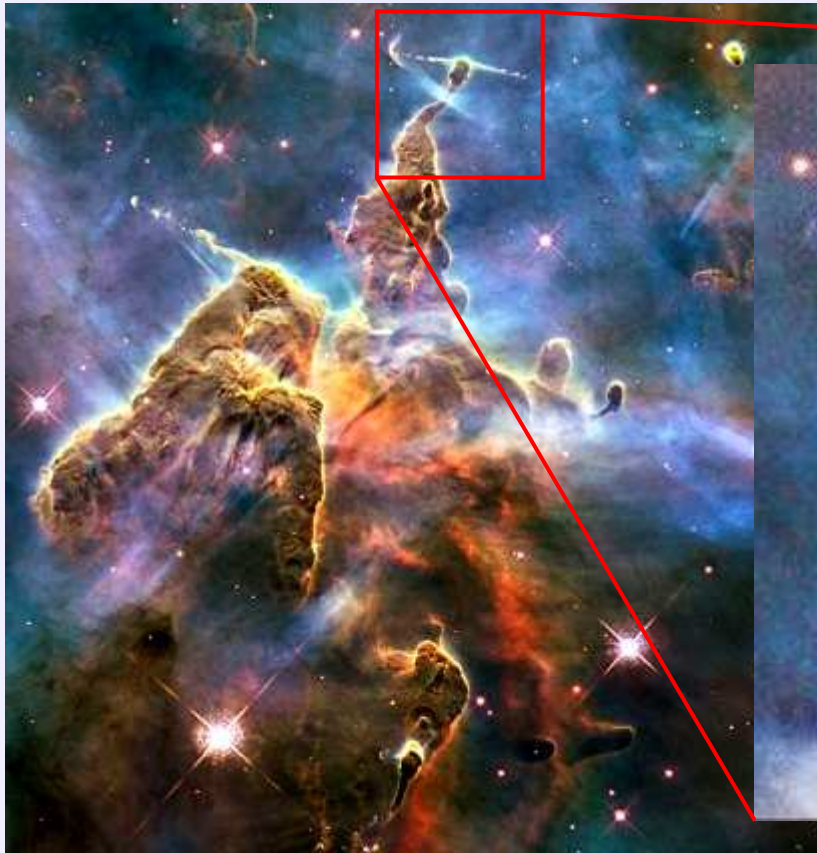
Plan to measure the half-life and neutron capture cross section of ^{53}Mn

Rugard Dressler

Outline

- Introduction
- MeaNCoRN
- Plans for ^{53}Mn
- ^{63}Ni as benchmark
- Summary

Formation of Stars



combined visible and near infrared pictures



star-forming pillar with a jet (HH 901)

Carina nebula: birthplace of stars

Hubble Space Telescope: <http://hubblesite.org>

Cosmogenic Radio-Nuclides (CoRN)

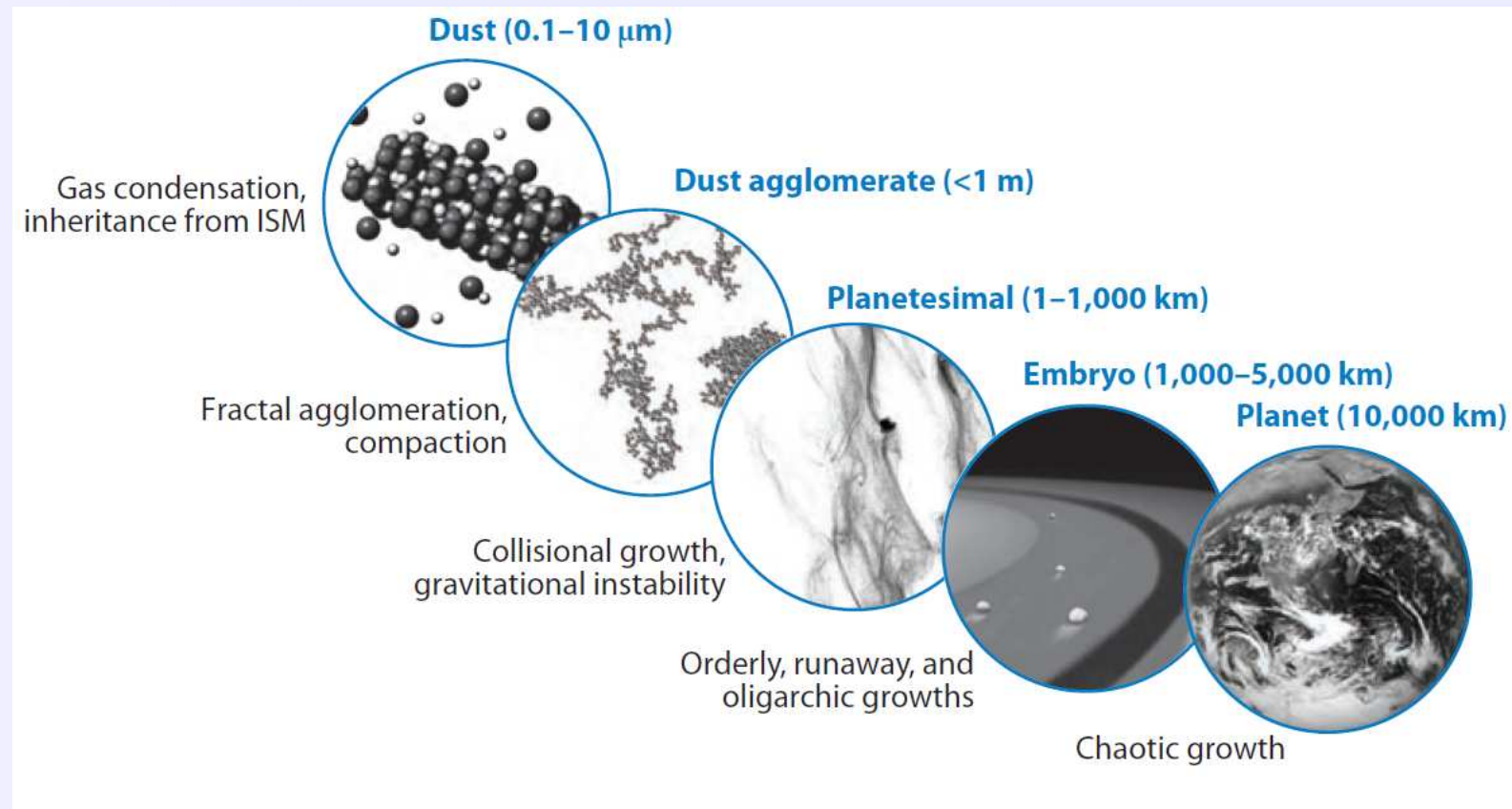
■ Produced

- ☐ *via neutron capture reaction*
- ☐ *during explosive phases of star evolution*
- ☐ *in spallation reactions with high energetic cosmic rays*

■ Present in the Solar System

- ☐ *as constitute of molecular cloud formed our Sun
e.g. found in meteorites*
- ☐ *injected continuously to „present days“ from super
novae
e.g. ^{60}Fe*
- ☐ *continuously produced at Earth from cosmic rays
e.g. ^7Be , ^{14}C , ^{53}Mn*

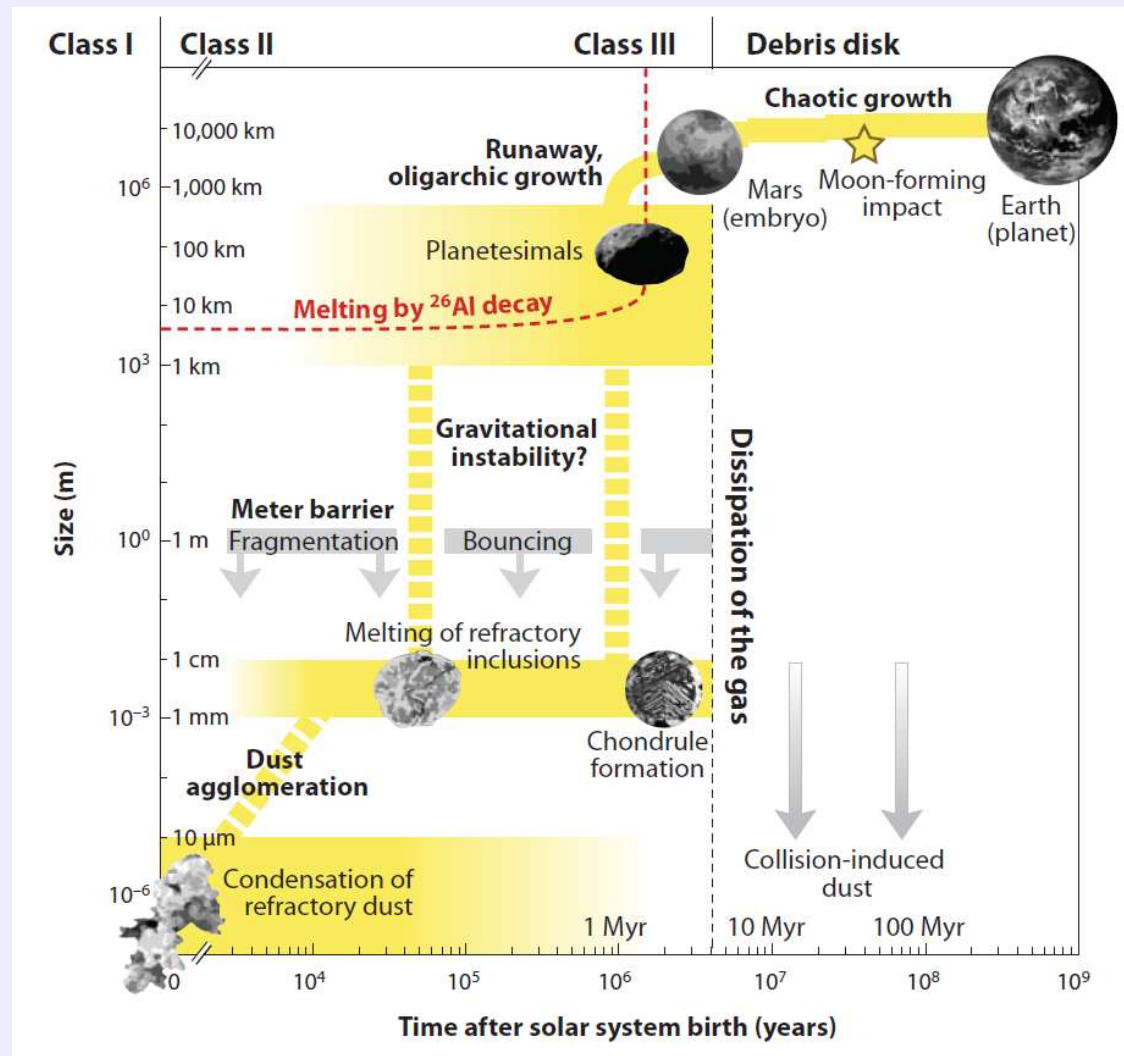
Formation of Planets



Evolution of matter in the inner part of the solar protoplanetary disk

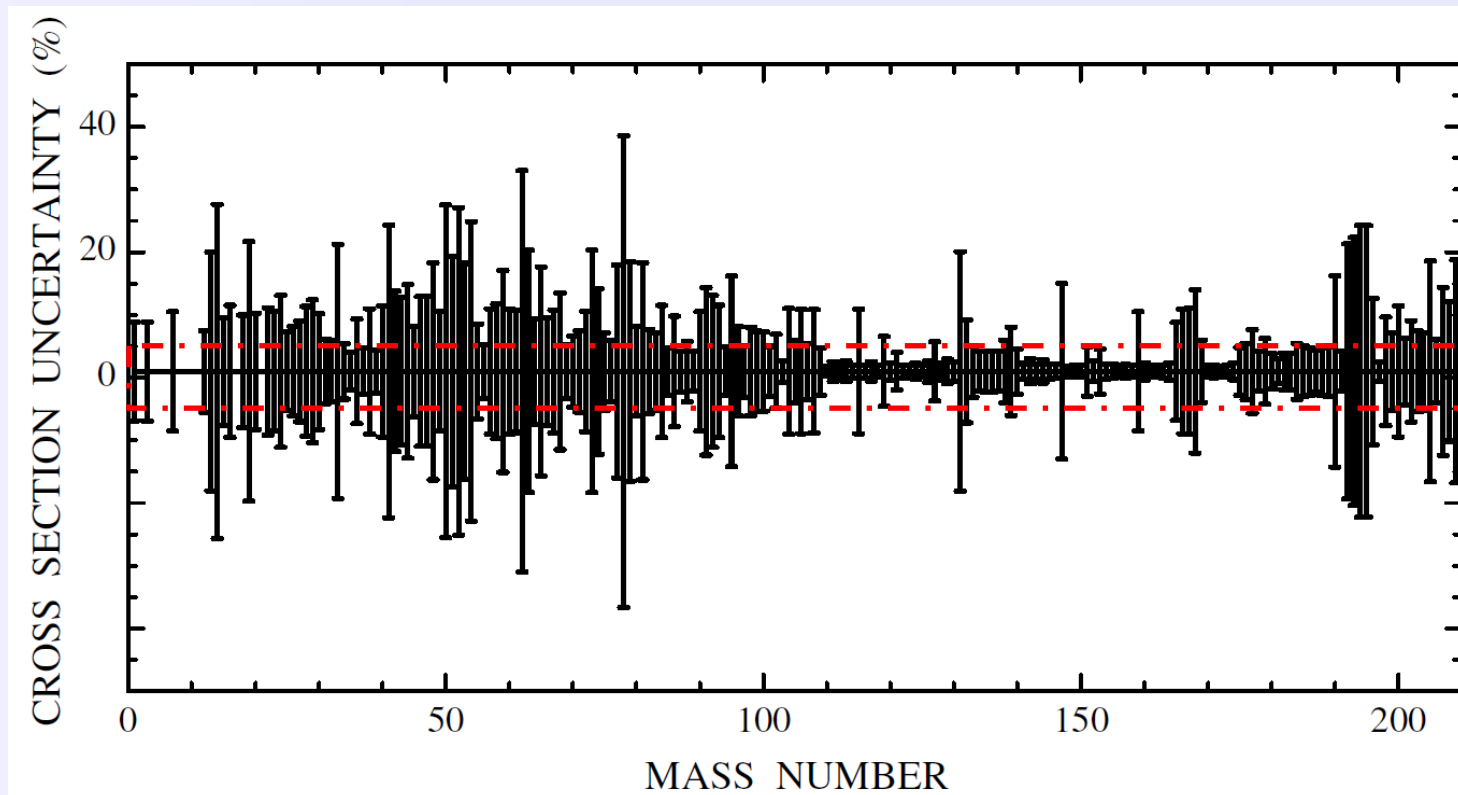
N. Dauphas, M. Chaussidon: Annu. Rev. Earth Planet. Sci. **39** (2011) 351

Role of CoRN in Planet Forming



N. Dauphas, M. Chaussidon: Annu. Rev. Earth Planet. Sci. **39** (2011) 351

Neutron Capture Cross-Sections at Stellar Neutron Energies



Short Lived Cosmogenic Radio-Nuclides

$t_{1/2} < 100 \text{ Ma}$

parent nuclide	half-life (Ma)	daughter nuclide	estim. init. abundance	Reference	at PSI
^{53}Mn	3.74	^{53}Cr	$^{53}\text{Mn} / ^{55}\text{Mn}$ $= 6.28 \cdot 10^{-6}$	Birck & Allègre 1985, Shukolyukov & Lugmair 2006, Trinquier et al. 2008	✓
^{10}Be	1.387	^{10}B	$^{10}\text{Be} / ^9\text{Be}$ $= 7.5 \cdot 10^{-4}$	McKeegan et al. 2000, Chaussidon & Gounelle 2006, 2007	✓
^{26}Al	0.717	^{26}Mg	$^{26}\text{Al} / ^{27}\text{Al}$ $= 5.23 \cdot 10^{-5}$	Jacobsen et al. 2008, Lee et al. 1976	✓
^{60}Fe	2.62	^{60}Ni	$^{60}\text{Fe} / ^{56}\text{Fe}$ $= 5.8 \cdot 10^{-9}$	Quitt'e et al. 2010, Shukolyukov & Lugmair 1993, Tang & Dauphas 2011	✓
^{36}Cl	0.301	^{36}S (~2%) ^{36}Ar (~98%)	$^{36}\text{Cl} / ^{35}\text{Cl}$ $> 17.2 \cdot 10^{-6}$	Jacobsen et al. 2009, Lin et al. 2005	✓
^{41}Ca	0.102	^{41}K	$^{41}\text{Ca} / ^{40}\text{Ca}$ $= 1.41 \cdot 10^{-8}$	Srinivasan et al. 1994, 1996	✓
^7Be	$1.46 \cdot 10^{-7}$	^7Li	$^7\text{Be} / ^9\text{Be}$ $= 6.1 \cdot 10^{-3}$	Chaussidon et al. 2006	✓

adapted from N. Dauphas, M. Chaussidon: Annu. Rev. Earth Planet. Sci. **39** (2011) 351

Summary

Ni-4

Mn-

Mean

Introduction

parent nuclide	half-life (Ma)	daughter nuclide	estim. init. abundance	Reference	at PSI
^{146}Sm	103	^{142}Nd	$^{146}\text{Sm} / ^{144}\text{Sm}$ $= 8.4 \cdot 10^{-3}$	Boyet et al. 2010, Lugmair & Galer 1992, Prinzhofer et al. 1992	☐
^{92}Nb	34.7	^{92}Zr	$^{92}\text{Nb} / ^{93}\text{Nb}$ $= 1.6 \cdot 10^{-5}$	Harper 1996, Schönbächler et al. 2002	?
^{129}I	15.7	^{129}Xe	$^{129}\text{I} / ^{127}\text{I}$ $= 1.19 \cdot 10^{-4}$	Brazzle et al. 1999, Jeffery & Reynolds 1961	?
^{182}Hf	8.9	^{182}W	$^{182}\text{Hf} / ^{180}\text{Hf}$ $= 9.72 \cdot 10^{-5}$	Burkhardt et al. 2008, Kleine et al. 2002, Yin et al. 2002	☐
^{107}Pd	6.5	^{107}Ag	$^{107}\text{Pd} / ^{108}\text{Pd}$ $= 5.9 \cdot 10^{-5}$	Chen & Wasserburg 1996, Schönbächler et al. 2008	?
^{205}Pb	15.1	^{205}Tl	$^{205}\text{Pb} / ^{204}\text{Pb}$ $= 1.0 \cdot 10^{-3}$	Baker et al. 2010	☐
^{135}Cs	2.3	^{135}Ba	$^{135}\text{Cs} / ^{133}\text{Cs}$ $= 4.8 \cdot 10^{-4}$	Hidaka et al. 2001	?
^{97}Tc	4.21	^{97}Mo	$^{97}\text{Tc} / ^{92}\text{Mo}$ $< 3 \cdot 10^{-6}$	Dauphas et al. 2002	?
^{98}Tc	4.2	^{98}Ru	$^{98}\text{Tc} / ^{96}\text{Ru}$ $< 2 \cdot 10^{-5}$	Becker & Walker 2003	?
^{126}Sn	0.23	^{126}Te	$^{126}\text{Sn} / ^{124}\text{Sn}$ $< 7.7 \cdot 10^{-5}$	Fehr et al. 2006	?

Summary

Ni-4

Mn-

Mean

Introduction

MeaNCoRN

Measurement of neutron capture cross sections
and determination of half-life of
short-lived cosmogenic radio-nuclides

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Summ

Ni-4

Mn-

MeaNCoRN

ction

Subjects of MeaNCoRN

■ Neutron Capture Cross Section

- ☐ *thermal neutron: NAA, PNA at PSI*
- ☐ *cold neutrons: BOA, ICON at PSI; FRM II in Munich*
- ☐ *ultra cold neutrons: UCN at PSI*
- ☐ *neutrons at cosmic energies: FRANZ at SGC Frankfurt*

■ Half-Life

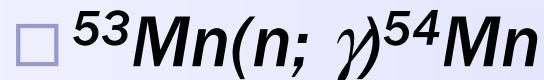
- ☐ ^{60}Fe , ^{63}Ni
- ☐ ^{10}Be , ^{53}Mn
- ☐ ^{26}Al , ^{32}Si , ^{59}Ni
- ☐ ^{41}Ca , ^{79}Se , ^{126}Sn , ...

■ Financial & Manpower Support

- ☐ *SNSF founding (rejected)*
- ☐ *PSI Cross Project*
- ☐ *??? other sources ???*

^{53}Mn

■ Capture Cross Section



- ^{54}Mn well known reference nuclide
- measurable with high sensitivity

■ Half-Life

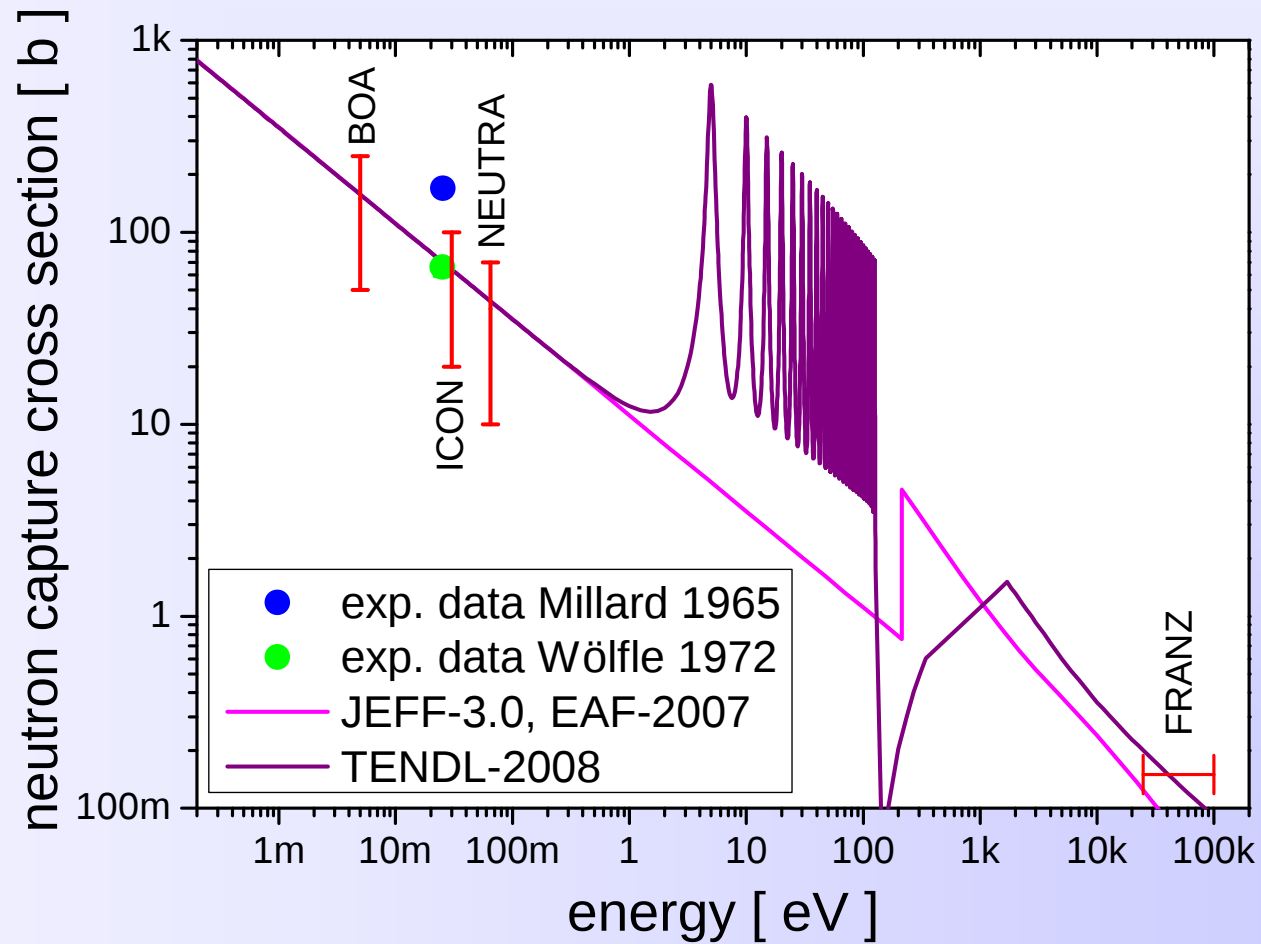


- used to date events in solar system (20 Ma)



- used to determine the terrestrial exposure time
Dry Valleys, Southern Victoria Land, and Antarctica

Neutron Capture Cross-Section ^{53}Mn

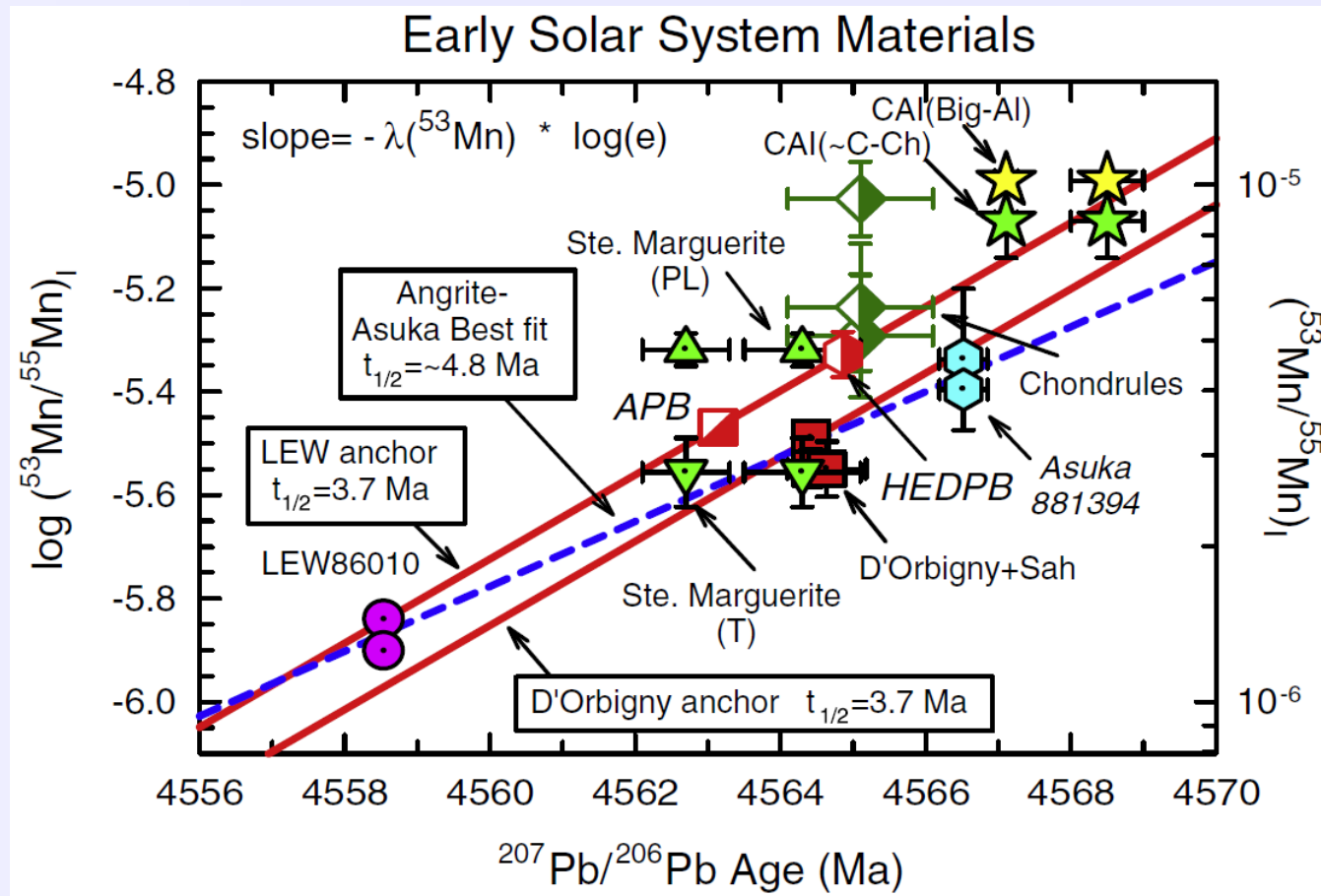


History of ^{53}Mn Half-Life Measurements

Author (Year)	Method	$t_{1/2}$ [Ma]
Wilkinson, et al. (1955)	Compared nuclear reaction yield with ^{54}Mn	0.00014
Sheline, et al. (1957)	Calculated nuclear reaction yield	~2
Kaye, et al. (1965)	Spallation yield of meteorites	1.9 ± 0.5
Hohlfelder (1969)	Mass spectrometry (MS) of meteorites	10.8 ± 4.5
Matsuda, et al. (1971)	MS of 730 MeV proton activation products	2.9 ± 1.2
Honda, et al. (1971)	MS of artificial and meteoritic samples	3.7 ± 0.37
Wölfle, et al. (1972)	Neutron activation of meteoritic samples	3.9 ± 0.6
Heimann, et al. (1974)	decay of meteoritic ^{53}Mn	3.85 ± 0.4

Dating Problem

$^{53}\text{Mn}/^{55}\text{Mn}$ vs. $^{207}\text{Pb}/^{206}\text{Pb}$





Fe samples of STIP
see talk Maruta Bunka

total material ~ 60 g steel

radio nuclides available

^{26}Al 300 Bq $\sim 10^{16}$ atoms
 ^{44}Ti 100 MBq $\approx 8.4 \cdot 10^{17}$ atoms
 ^{54}Mn 70 MBq
 (^{53}Mn $\sim 10^{19}$ atoms)
 ^{60}Co 70 MBq

CoRN Storehouses



Cu beam dump
see talk Marin Ayrarov

total material ~ 500 g copper

radio nuclides inventory

^{26}Al 7 kBq $\approx 2.3 \cdot 10^{17}$ atoms
 ^{32}Si 10 MBq $\approx 7.8 \cdot 10^{16}$ atoms
 ^{44}Ti 100 MBq $\approx 2.8 \cdot 10^{17}$ atoms
 ^{53}Mn 500 kBq $\approx 8.4 \cdot 10^{19}$ atoms
 ^{59}Ni 8 MBq $\approx 2.7 \cdot 10^{19}$ atoms
 ^{60}Fe 5 kBq $\approx 5.9 \cdot 10^{17}$ atoms
 ^{60}Co 5 GBq

CoRN Storehouses

STIP samples

pro

- ☐ *low ^{60}Co content*
- ☐ *separated from matrix*
- ☐ *purified final product*

cons

- ☐ *high amount of stable isotopes*
- ☐ *storage time ~12 a
EoB Dec 1999*
- ☐ *high ^{54}Mn activity*

Cu beam dump

pro

- ☐ *storage time ~20 a
EoB Dec 1992*
- ☐ *low ^{54}Mn activity*
- ☐ *non carrier added*

cons

- ☐ *high activity of ^{60}Co*
- ☐ *not separated from Cu*
- ☐ *not purified*

Half-Life Measurement

$$t_{1/2} > 50 \text{ a}$$

Determination of sample activity

- α - / β - / γ -Spectroscopy or Liquid Scintillation Counting

Determination of number of atoms

- AMS / ICP-MS / HI-ERD

- ☐ $^{10}\text{Be } t_{1/2} = 1.388 \pm 0.018 \text{ Ma}$
G. Korschinek, et al.: Nucl. Instr. and Methods B 268 (2010) 187
- ☐ $^{10}\text{Be } t_{1/2} = 1.386 \pm 0.016 \text{ Ma}$
J. Chmeleff, et al.: Nucl. Instr. and Methods B 268 (2010) 192
- ☐ $^{79}\text{Se } t_{1/2} = 0.327 \pm 0.008 \text{ Ma}$
G.Jörg, et al.: Applied Radiation and Isotopes 68 (2010) 2339
- ☐ $^{60}\text{Fe } t_{1/2} = 2.62 \pm 0.04 \text{ Ma}$
G. Rügeli, et al.: Phys. Rev. Lett. 103 (2009) 072502

^{63}Ni Half-Life Measurements at PSI

^{63}Ni pure β^- emitter

$$E_{\text{max}} = 66.9 \text{ keV}$$

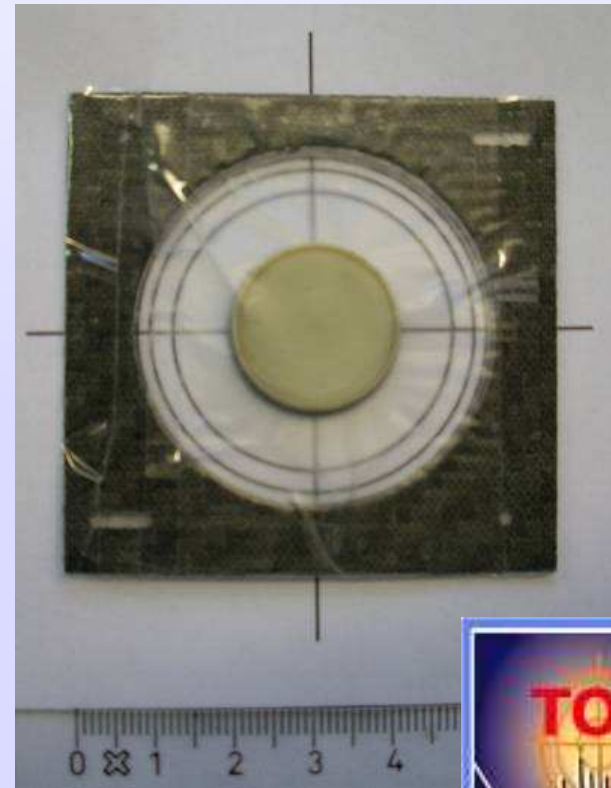
^{63}Ni target for n_ToF

- production

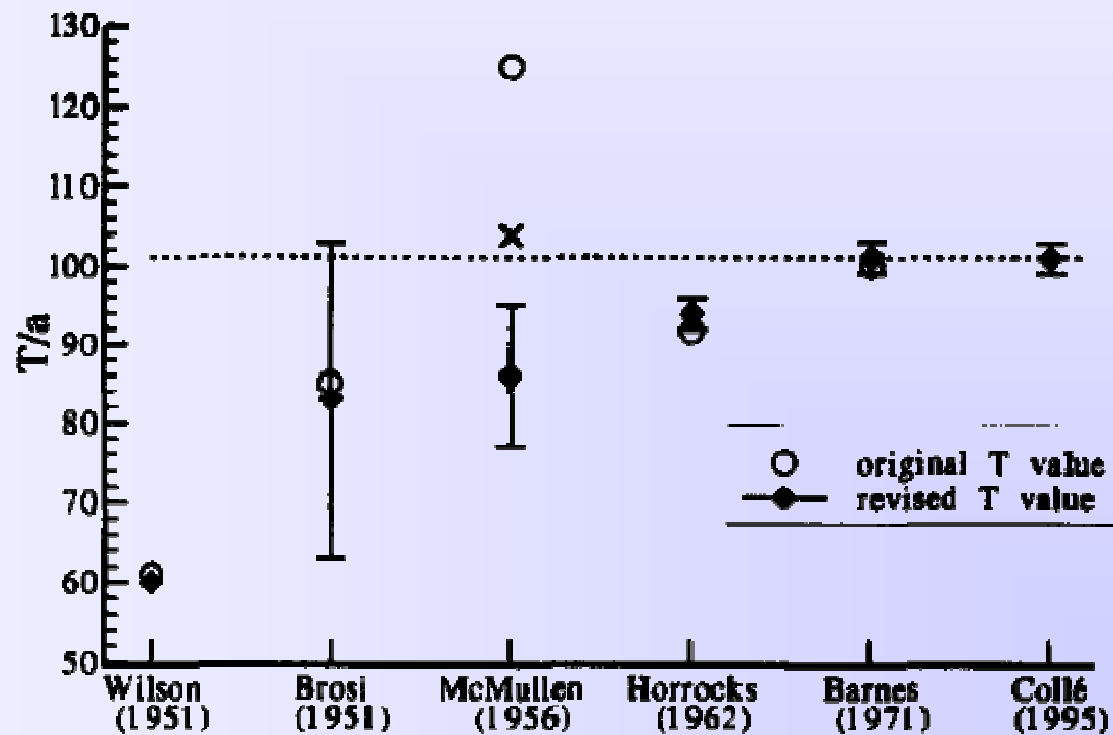
- ☐ 988 mg enriched ^{62}Ni
metal pellets neutron activated
- ☐ 109 mg ^{63}Ni
 $\approx 10^{21}$ atoms $\approx 230 \text{ GBq}$
- ☐ separation of ^{63}Cu

- about 0.5% left as waste
 $\approx 5 \cdot 10^{18}$ atoms $\approx 1 \text{ GBq}$

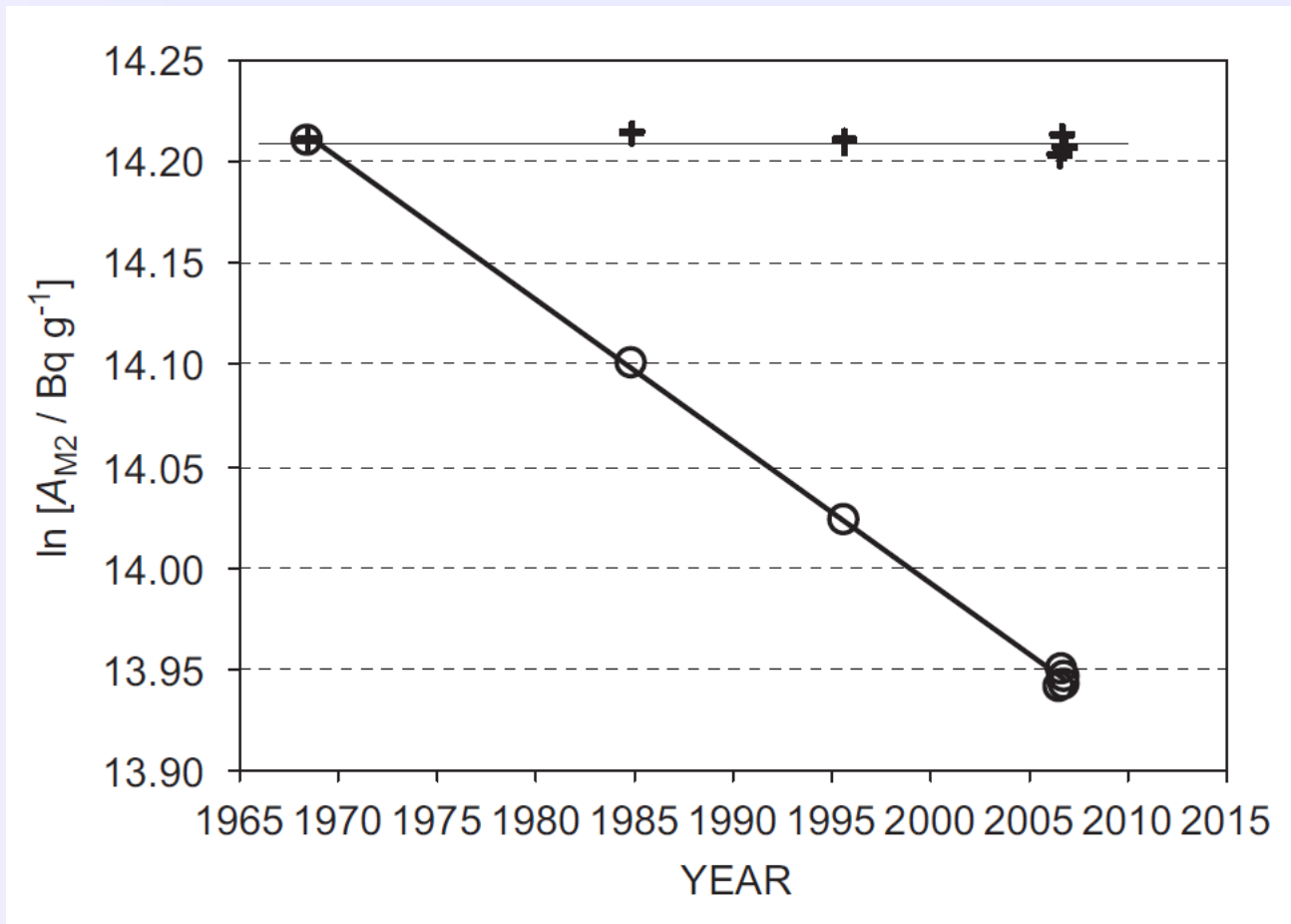
- can be used for
other measurements



History of ^{63}Ni Half-Life Measurements



^{63}Ni Half-Life and Standardization



Why additional Measurements?

- **Collè et al. 1996:**
new value based on the revised value from Barnes et. Al
own ones using decay measurements with LSC technique
3 data points over 27 years: 101.1 ± 1.4 a
 - **Collè et al. 2008:**
new re-standardization of the used NIST solutions
additional data points for the decay measurement
4 points over about 40 years: 101.2 ± 1.5 a
1. Additional measurements using other techniques would be useful to support the result of Collè.
 2. ^{63}Ni is an excellent play ground to prove the precision of our used instruments and methods.

Half-Life Measurements at PSI

1. Approach

$$t_{1/2} = \Delta t \cdot \ln 2 / \ln(N_0/N)$$

- ICP-MS measurements:

1. isotopic ratio before purification: $(^{63}\text{Ni} + ^{63}\text{Cu}) / ^{62}\text{Ni} \rightarrow N_0$

2. isotopic ratio after purification: $^{63}\text{Ni} / ^{62}\text{Ni} \rightarrow N$

- Time from EoB to separation: $???$ $\rightarrow \Delta t$

sample	$^{63}/^{62}\text{Ni}$ (K)	uncertainty	$^{63}/^{62}\text{Ni}$ (L)	uncertainty
after dissolving	0.145366	0.000037	0.124784	0.000266
after CuS separation	0.125098	0.000010	0.107667	0.000013
after purification	0.122395	0.000289	0.107846	0.000169
N_0/N	1.1877	0.0028	1.1590	0.0025
Δt	? (25 a)		? (21.5 a)	
$t_{1/2}$	(101.2 a)		(101.2 a)	

Planned Half-Life Measurements

2. Approach

$$t_{1/2} = N \cdot \ln 2 / A$$

- **Isotope dilution**

1. Ad 12 mg ^{nat}Cu
2. Separation of Cu with H₂S (several times)
3. final purification
4. stock solution for measurements

- **ICP-MS measurement:** → N (pending)

interference correction of mass 63

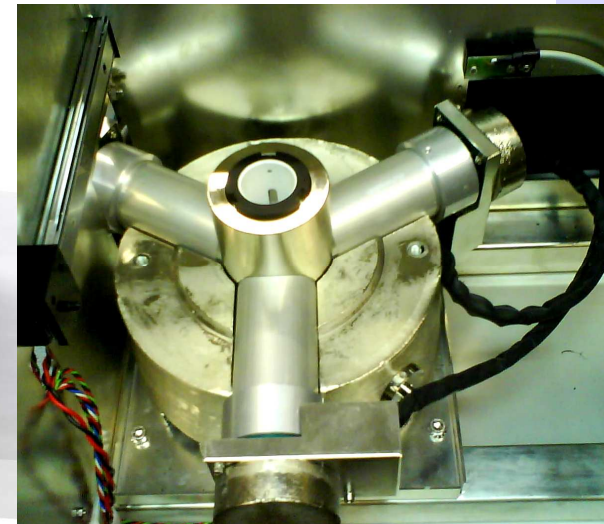
comparison with certificated Ni reference material

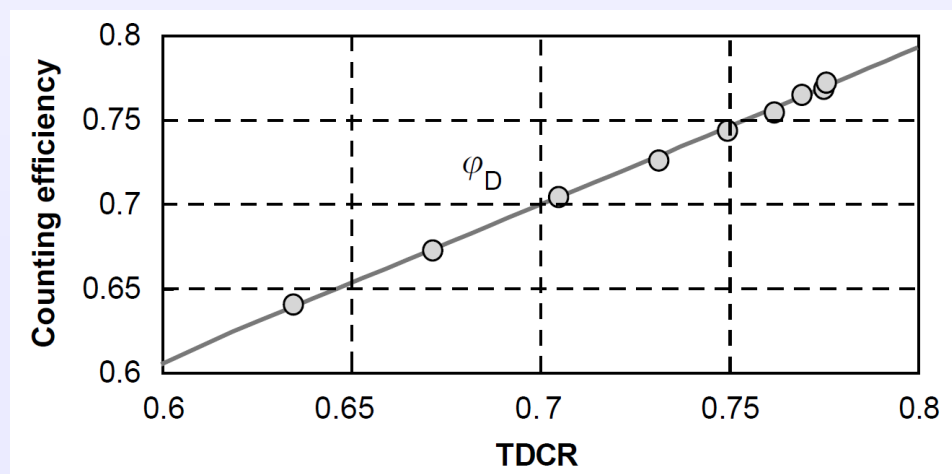
- **LSC measurements:** → A = 520.7 ± 0.7 Bq/g

calibration of quench effects

comparison with certificated standards of other β⁻ emitters

Liquid Scintillation Counter Hidex 300 SL





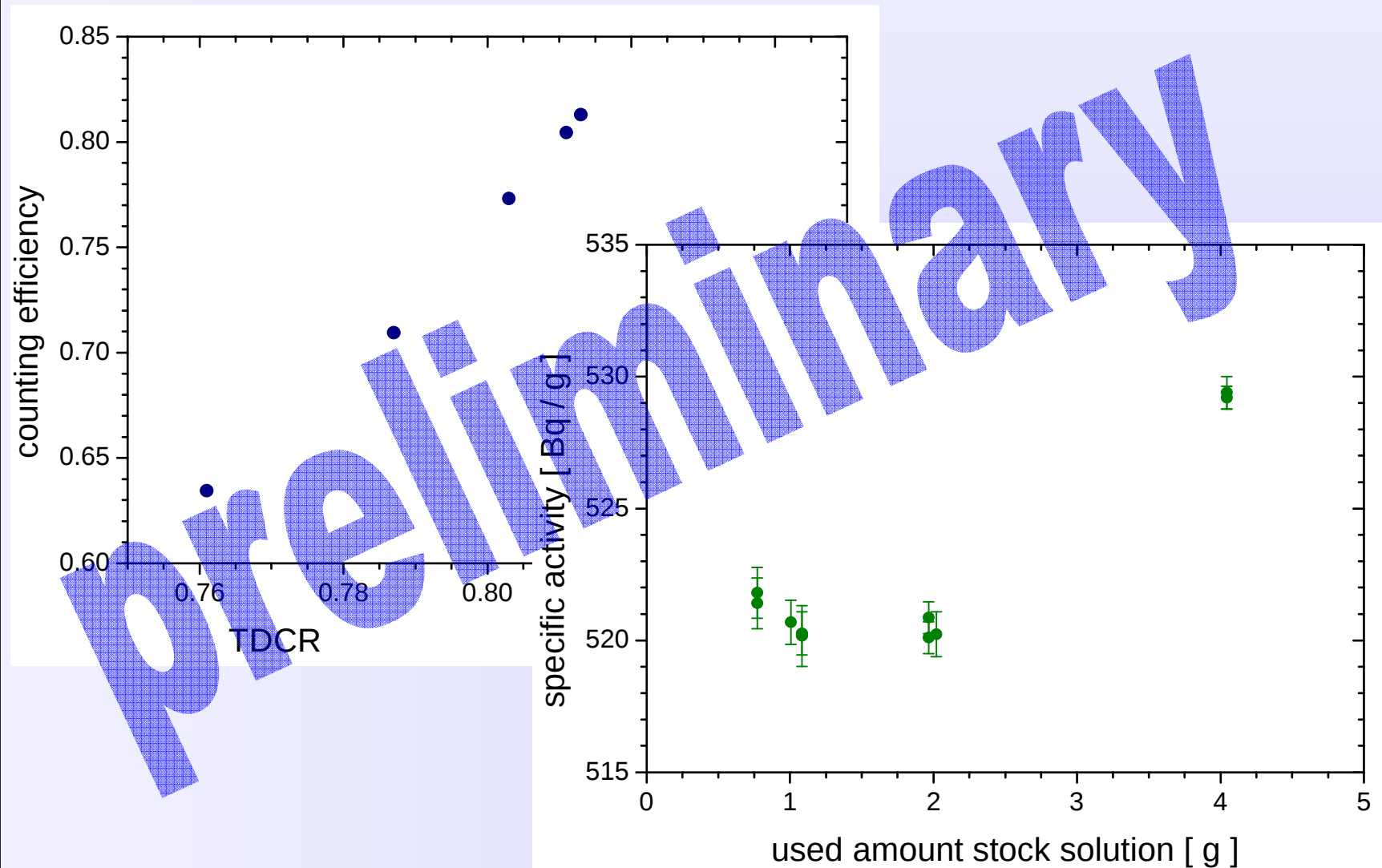
counting efficiency of a ^{63}Ni source
in Ultima Gold liquid scintillator

R. Broda: Applied Radiation and Isotopes 58 (2003) 585

Sample	TDCR	Eff _D [s ⁻¹ Bq ⁻¹]	Dev
10 ml UG ^{AB}	0.83	0.805	-0.030
10 ml UG TM	0.84	0.820	-0.027
Hisafe	0.83	0.800	-0.035

results obtained for ^{63}Ni (10ml UGAB) using TDCRB-02
I.A. Kharitonov, T.I.Shilnikova: (2009)

First Results of LSC



Summary

- ^{26}Al , ^{32}Si , ^{44}Ti , ^{53}Mn , ^{59}Ni , ^{60}Fe
available at PSI (10^{16} – 10^{20} atoms)
- MeaNCoRN
 - ☐ *experiments at PSI*
- ^{53}Mn
 - ☐ *up to 10^{20} available*
 - ☐ *re-measurment of half-life*
- ^{63}Ni
 - ☐ *model isotope for performance tests*
 - ☐ *re-measurment of half-life (on going)*