

PSI Master School 2017

Introducing photons, neutrons and
muons for materials characterization

Lecture 8: Neutron diffraction

Importance of nuclear neutron diffraction

- Structure determines physical properties
- Detailed knowledge of atomic positions and lattice symmetry is essential
- Structure is measured with elastic scattering: there is no energy transfer

Bragg peak scattering in Bravais lattice

$$\left(\frac{d\sigma}{d\Omega}\right)_{\text{coh}} = \langle b \rangle^2 \sum_{j,j'} e^{-i\mathbf{Q} \cdot (\mathbf{R}_{j'} - \mathbf{R}_j)} \quad \mathbf{Q} = \mathbf{k}_f - \mathbf{k}_i$$

$\mathbf{r} = \mathbf{R}_{j'} - \mathbf{R}_j$ and multiply with number of unit cell N_0
(assumption: only one type of atom with scattering length b on Bravais lattice)

$$\frac{d\sigma}{d\Omega} = N_0 \langle b \rangle^2 \sum_{\mathbf{r}} e^{i\mathbf{Q} \cdot \mathbf{r}}$$

Perform sum over \mathbf{r} :

$$\sum_{\mathbf{r}} e^{i\mathbf{Q} \cdot \mathbf{r}} = \frac{(2\pi)^3}{v_0} \sum_{\boldsymbol{\tau}} \delta(\mathbf{Q} - \boldsymbol{\tau})$$

$$\frac{d\sigma}{d\Omega} = N_0 \frac{(2\pi)^3}{v_0} \langle b \rangle^2 \sum_{\boldsymbol{\tau}} \delta(\mathbf{Q} - \boldsymbol{\tau})$$

Reminder: reciprocal lattice

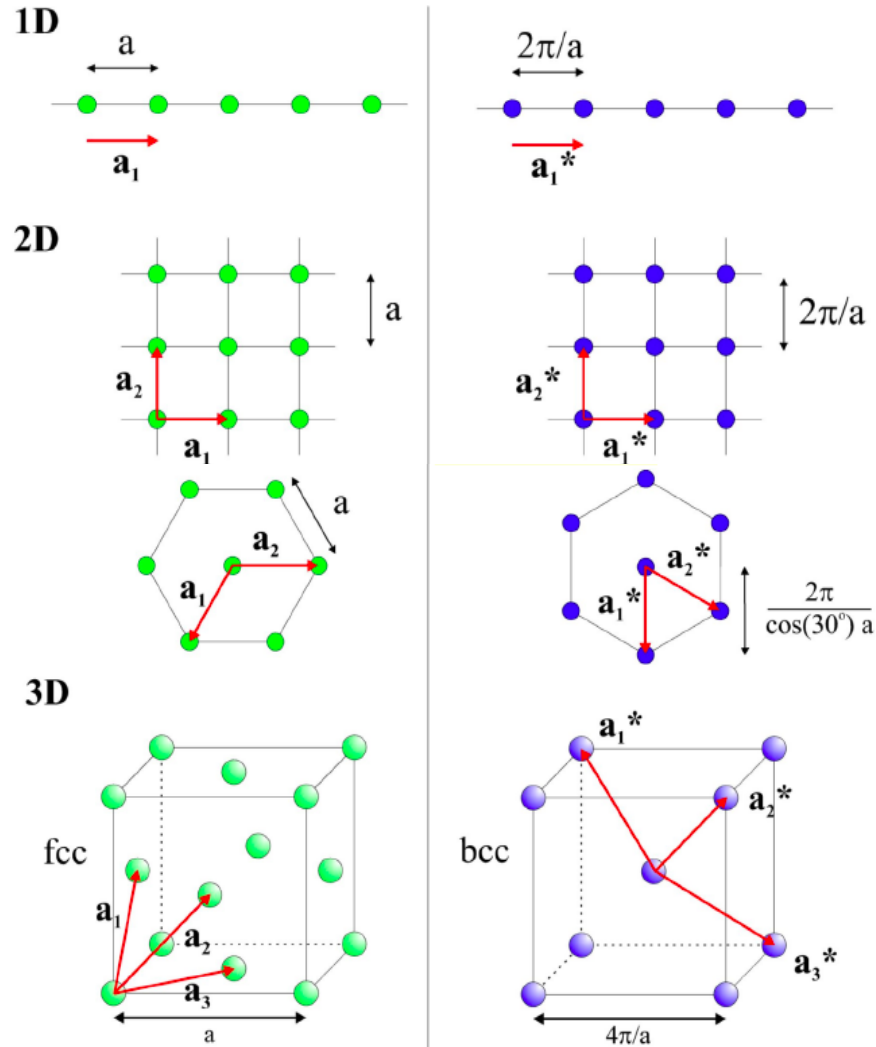
Lattice for “wave-vector space”

Connected with real space via
Fourier transform

$$\mathbf{a}^* = 2\pi \frac{\mathbf{b} \times \mathbf{c}}{\mathbf{a} \cdot (\mathbf{b} \times \mathbf{c})}$$

$$\mathbf{b}^* = 2\pi \frac{\mathbf{c} \times \mathbf{a}}{\mathbf{a} \cdot (\mathbf{b} \times \mathbf{c})}$$

$$\mathbf{c}^* = 2\pi \frac{\mathbf{a} \times \mathbf{b}}{\mathbf{a} \cdot (\mathbf{b} \times \mathbf{c})}$$



Bragg peak scattering in the non-Bravais case

$$\frac{d\sigma}{d\Omega} = \sum_{j,j'} b_j b_{j'} \langle e^{-i\mathbf{Q} \cdot \hat{\mathbf{R}}_{j'}} e^{i\mathbf{Q} \cdot \hat{\mathbf{R}}_j} \rangle.$$

For a non-Bravais lattice the atomic positions are given by the position of the unit cell (lattice vector) plus relative position of the atom within unit cell

$$\mathbf{R} = \mathbf{l}_j + \mathbf{d}_\alpha$$

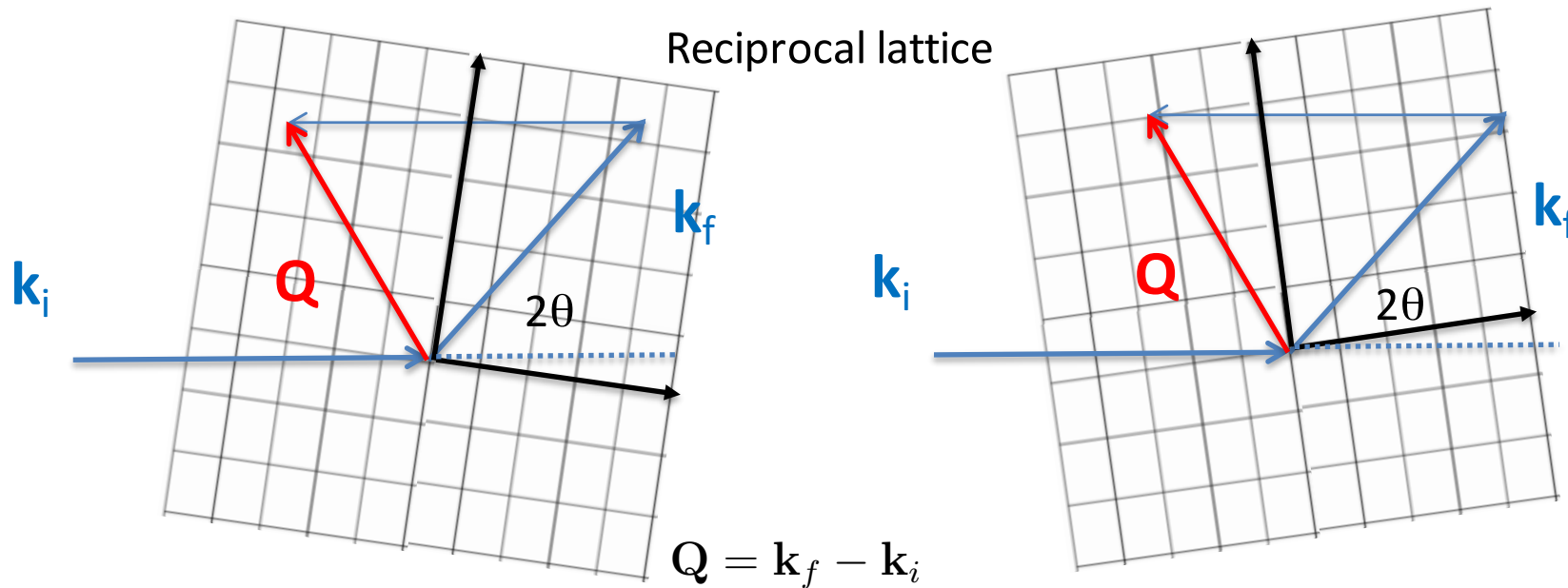
$$\frac{d\sigma}{d\Omega} = \sum_{j,j'} e^{i\mathbf{Q} \cdot (\mathbf{l}_j - \mathbf{l}_{j'})} \sum_{\alpha,\alpha'} b_\alpha b_{\alpha'} e^{i\mathbf{Q} \cdot (\mathbf{d}_\alpha - \mathbf{d}_{\alpha'})}.$$

$$\text{using} \quad \sum_{\alpha,\alpha'} b_\alpha b_{\alpha'} e^{i\mathbf{Q} \cdot (\mathbf{d}_\alpha - \mathbf{d}_{\alpha'})} = \left| \sum_{\mathbf{d}} b_{\mathbf{d}} e^{i\mathbf{Q} \cdot \mathbf{d}} \right|^2$$

$$\frac{d\sigma}{d\Omega} = N_0 \frac{(2\pi)^3}{v_0} \sum_{\boldsymbol{\tau}} |F_N(\boldsymbol{\tau})|^2 \delta(\mathbf{Q} - \boldsymbol{\tau})$$

$$F_N(\boldsymbol{\tau}) = \sum_{\mathbf{d}} b_{\mathbf{d}} e^{i\boldsymbol{\tau} \cdot \mathbf{d}}$$

Bragg scattering occurs in specific directions



Bragg condition: $Q = 2k_i \sin \Theta$
 $Q = \tau$

Crystal has to be aligned accurately to observed Bragg scattering

Simple diffractometer offers the possibility the sample and change toe scattering angle

How to look for a Bragg peak?

Debye-Waller factor

$$\frac{d\sigma}{d\Omega} = \sum_{j,j'} b_j b_{j'} \langle e^{-i\mathbf{Q} \cdot \hat{\mathbf{R}}_{j'}} e^{i\mathbf{Q} \cdot \hat{\mathbf{R}}_j} \rangle.$$

The atoms thermally move around their equilibrium positions. As a consequence the scattered intensity falls off at higher momentum transfer \mathbf{Q} , according to the so-called Debye-Waller factor

$$\text{For cubic material} \quad 2W(\mathbf{Q}) = 2W(Q) = \frac{1}{3} Q^2 \langle u^2 \rangle$$

$$\frac{d\sigma}{d\Omega} = N_0 \frac{(2\pi)^3}{v_0} e^{-2W(\mathbf{Q})} \sum_{\boldsymbol{\tau}} |F_N(\boldsymbol{\tau})|^2 \delta(\mathbf{Q} - \boldsymbol{\tau})$$

$$F_N(\boldsymbol{\tau}) = \sum_{\mathbf{d}} b_{\mathbf{d}} e^{i\boldsymbol{\tau} \cdot \mathbf{d}}$$

Summary nuclear Bragg scattering

$$\frac{d\sigma}{d\Omega} = N_0 \frac{(2\pi)^3}{v_0} e^{-2W(Q)} \sum_{\tau} |F_N(\tau)|^2 \delta(Q - \tau)$$

$$F_N(\tau) = \sum_d b_d e^{i\tau \cdot d}$$

Debye-Waller factor

Information of atomic
displacements from
equilibrium positions

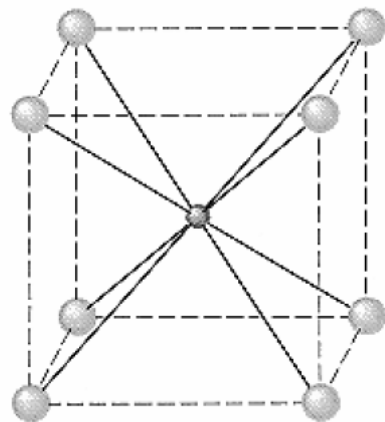
Intensities of Bragg peaks

Information about the
species and location of
atoms within unit cell

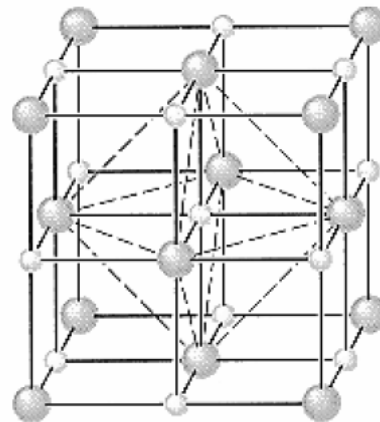
Location of Bragg peaks

Information about chemical
lattice (lattice parameters and
symmetry of space group)

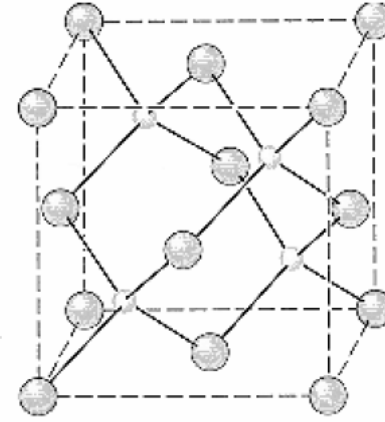
Excursion: symmetry



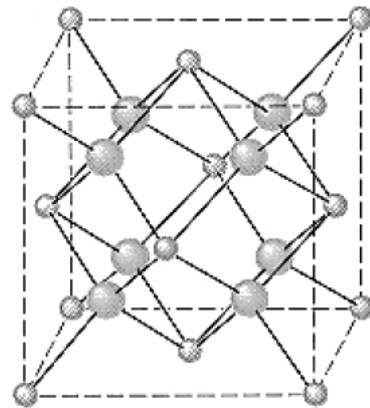
CsCl



NaCl

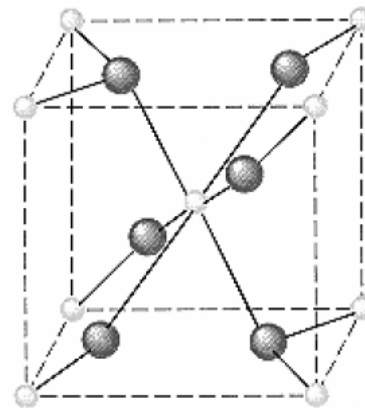


Zinc blende (cubic ZnS)



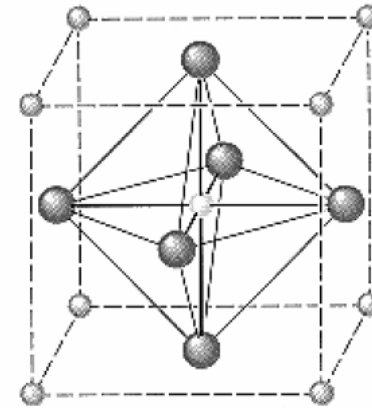
Fluorite (CaF₂)

● = Ca²⁺



Rutile (TiO₂)

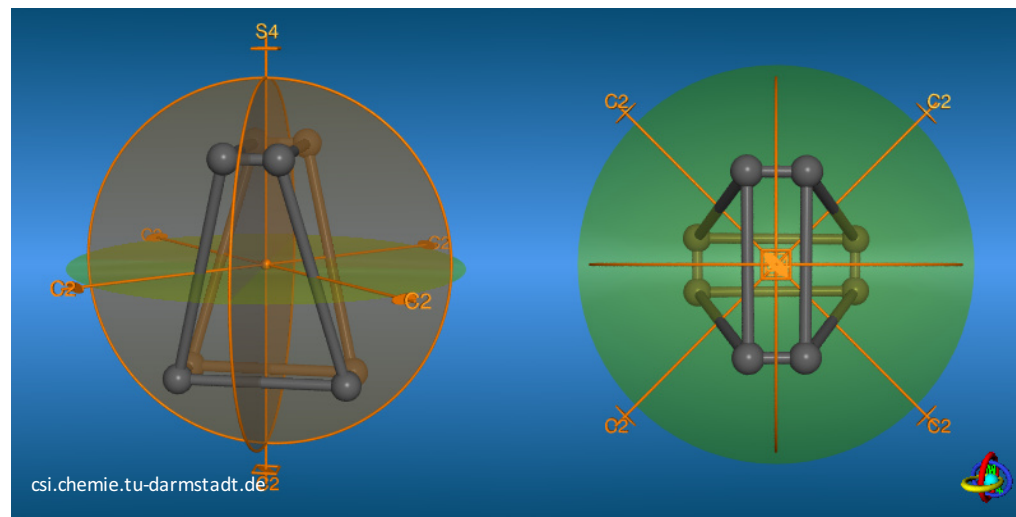
● = Ti^{IV}



Perovskite (CaTiO₃)

● = Ti^{IV} ● = Ca²⁺ ● = O²⁻

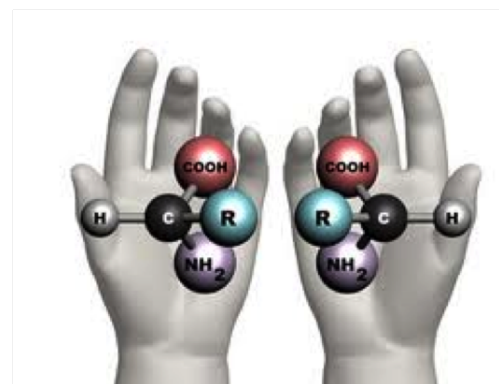
Point groups



Point groups determine local properties, such as magnetic moments, crystal fields etc

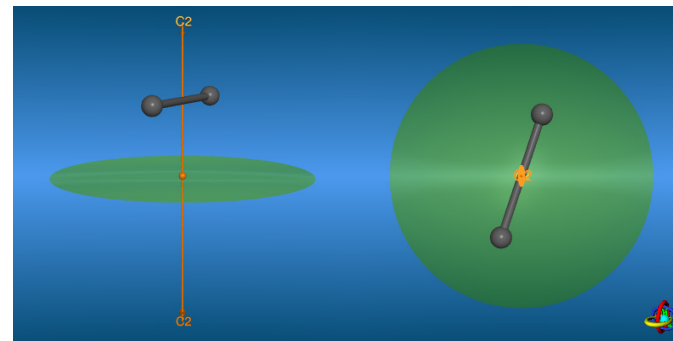
Basic symmetry operations:

- Rotations
- Mirror planes
- Inversion

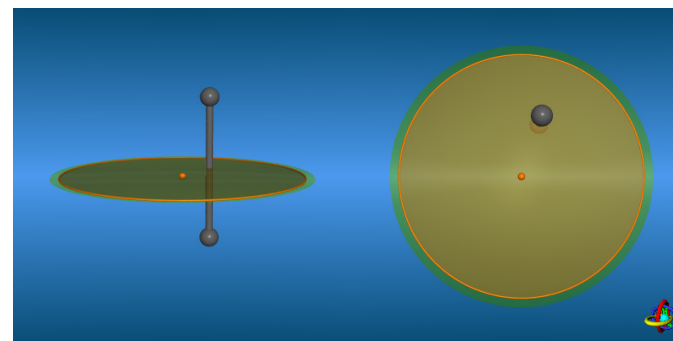


Basic symmetry operations

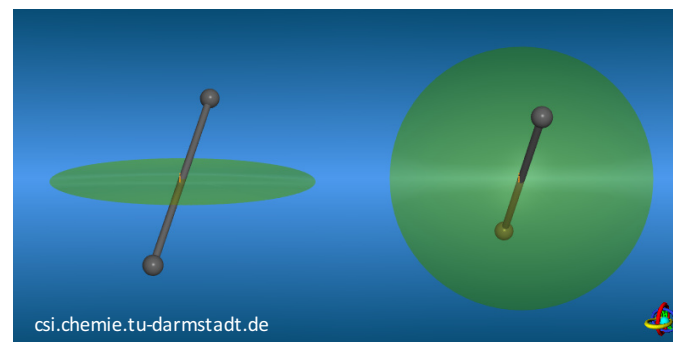
Rotation symmetry $\mathcal{R} = \begin{pmatrix} \cos \phi & -\sin \phi & 0 \\ \sin \phi & \cos \phi & 0 \\ 0 & 0 & 1 \end{pmatrix}$



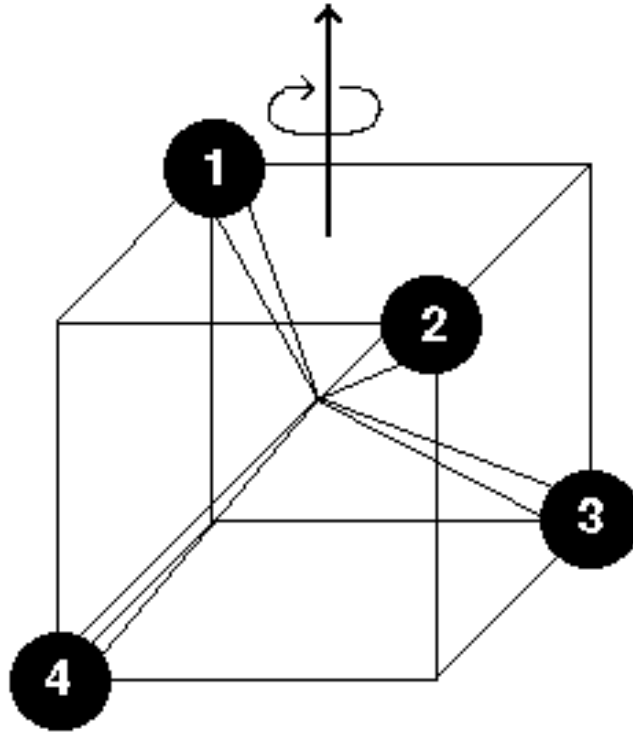
Mirror symmetry $\mathcal{R} = \begin{pmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & -1 \end{pmatrix}$



Inversion symmetry



Example of combined symmetry operations



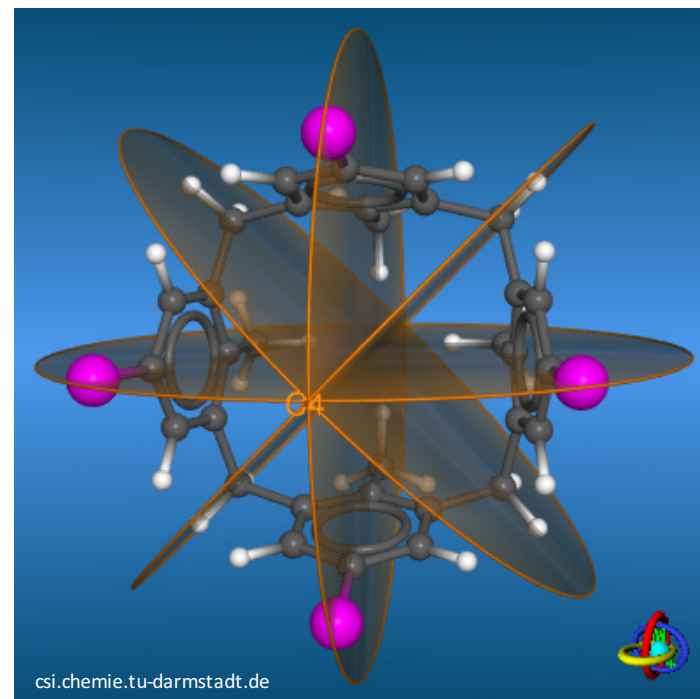
Symmetry operation S_4 : four-fold rotation followed by a mirror operation perpendicular to the four-fold axis

Example

Point group C_{4v}

- Two-fold axis
- Four-fold axis
- Four mirror planes

$4mm$	1	2_z	4_z	4_z^{-1}	m_{xz}	m_{yz}	m_{xx}	$m_{x\bar{x}}$
1	1	2_z	4_z	4_z^{-1}	m_{xz}	m_{yz}	m_{xx}	$m_{x\bar{x}}$
2_z	2_z	1	4_z^{-1}	4_z	m_{yz}	m_{xz}	$m_{x\bar{x}}$	m_{xx}
4_z	4_z	4_z^{-1}	2_z	1	m_{xx}	$m_{x\bar{x}}$	m_{yz}	m_{xz}
4_z^{-1}	4_z^{-1}	4_z	1	2_z	$m_{x\bar{x}}$	m_{xx}	m_{xz}	m_{yz}
m_{xz}	m_{xz}	m_{yz}	$m_{x\bar{x}}$	m_{xx}	1	2_z	4_z^{-1}	4_z
m_{yz}	m_{yz}	m_{xz}	m_{xx}	$m_{x\bar{x}}$	2_z	1	4_z	4_z^{-1}
m_{xx}	m_{xx}	$m_{x\bar{x}}$	m_{xz}	m_{yz}	4_z	4_z^{-1}	1	2_z
$m_{x\bar{x}}$	$m_{x\bar{x}}$	m_{xx}	m_{yz}	m_{xz}	4_z^{-1}	4_z	2_z	1



List of point groups

Crystal System	Number of Point Groups	Herman-Mauguin Point Group	Schoenflies Point Group
Triclinic	2	1, $\bar{1}$	C_1 , C_i
Monoclinic	3	2, m, 2/m	C_2 , C_s , C_{2h}
Orthorhombic	3	222, mm2, mmm	D_2 , C_{2v} , D_{2h}
Trigonal	5	3, $\bar{3}$, 32 3m, $\bar{3}m$	C_3 , S_6 , D_3 C_{3v} , D_{3d}
Hexagonal	7	6, $\bar{6}$, 6/m, 622, 6mm, 62m, 6mm	C_6 , C_{3h} , C_{6h} , D_6 C_{6v} , D_{3h} , D_{6h}
Tetragonal	7	4, $\bar{4}$, 4/m, 422, 4mm, $\bar{4}2m$ 4/mmm	C_4 , S_4 , C_{4h} , D_4 C_{4v} , D_{2d} , D_{4h}
Cubic	5	23, m3, 432 $\bar{4}32$, $m\bar{3}m$	T, T_h , O T_d , O_h

Space groups

point groups + lattice translations = space groups

$$(\mathcal{R}, \mathbf{t})(\mathbf{r}) = \mathcal{R}\mathbf{r} + \mathbf{t}$$

$$(\mathcal{R}_1, \mathbf{t}_1) \cdot (\mathcal{R}_2, \mathbf{t}_2) = (\mathcal{R}_1 \cdot \mathcal{R}_2, \mathcal{R}_1\mathbf{t}_1 + \mathbf{t}_2)$$

Crystal symmetry consists of

- Bravais lattices
- Choice of basis
- Translational symmetry

There are 230 space groups in three dimensions

→ International Tables for Crystallography (Hahn)

Space groups: international tables

Home

<http://it.iucr.org/>

International Tables for Crystallography

ISBN: 978-0-470-68575-4 doi: 10.1107/97809553602060000001

This is the home page for **International Tables**, the definitive resource and reference work for crystallography. The series consists of the following volumes:



Guided tour

Volume A Space-group symmetry

2016 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Volume A1 Symmetry relations between space groups

2011 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Volume B Reciprocal space

2010 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Volume C Mathematical, physical and chemical tables

2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Volume D Physical properties of crystals

2013 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Volume E Subperiodic groups

2010 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Volume F Crystallography of biological macromolecules

2012 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Volume G Definition and exchange of crystallographic data

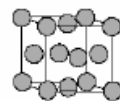
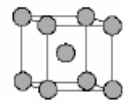
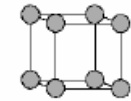
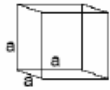
2006 Edition | [Contents](#) | [Sample pages](#) | [Indexes](#) |

Reminder Bravais lattice

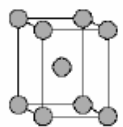
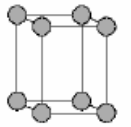
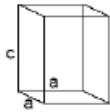
7 Crystal systems

14 Bravais Lattices

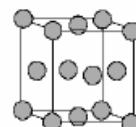
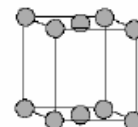
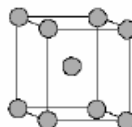
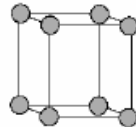
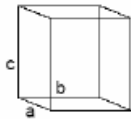
cubic
 $a=b=c$
 $\alpha=\beta=\gamma=90^\circ$



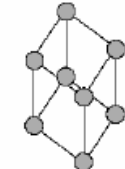
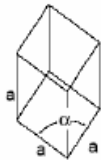
tetragonal
 $a=b \neq c$
 $\alpha=\beta=\gamma=90^\circ$



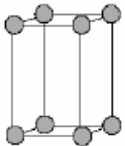
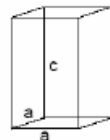
orthorhombic
 $a \neq b \neq c$
 $\alpha=\beta=\gamma=90^\circ$



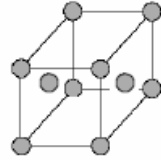
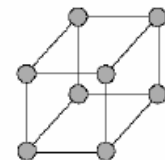
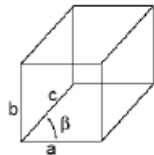
rhombohedral
 $a=b=c$
 $\alpha=\beta=\gamma \neq 90^\circ$



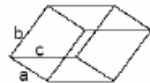
hexagonal
 $a=b \neq c$
 $\alpha=\beta=90^\circ$
 $\gamma=120^\circ$



monoclinic
 $a \neq b \neq c$
 $\alpha=\gamma=90^\circ \neq \beta$



triclinic
 $a \neq b \neq c$
 $\alpha \neq \beta \neq \gamma \neq 90^\circ$



centering	symbol	centering vector(s)	lattice points per unit cell
face centered	F	$\frac{1}{2}(\mathbf{b} + \mathbf{c})$ $\frac{1}{2}(\mathbf{a} + \mathbf{c})$ $\frac{1}{2}(\mathbf{a} + \mathbf{b})$	4
body centered	I	$\frac{1}{2}(\mathbf{a} + \mathbf{b} + \mathbf{c})$	2
base centered	A	$\frac{1}{2}(\mathbf{b} + \mathbf{c})$	2
base centered	B	$\frac{1}{2}(\mathbf{a} + \mathbf{c})$	2
base centered	C	$\frac{1}{2}(\mathbf{a} + \mathbf{b})$	2

Important properties of a crystal structure

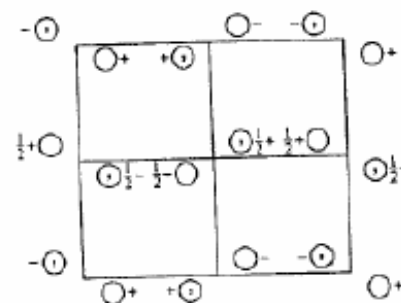
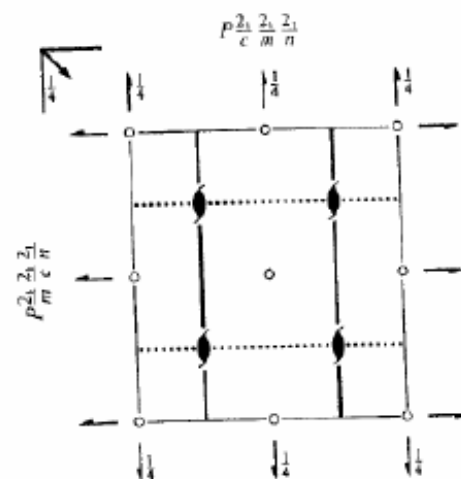
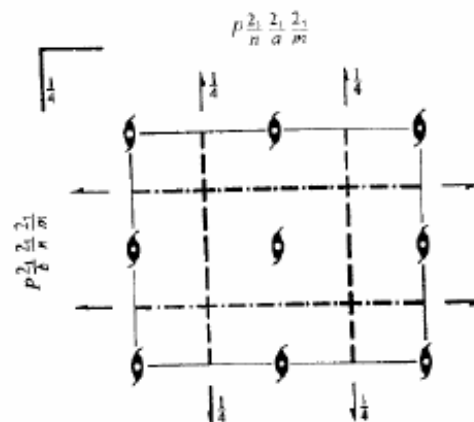
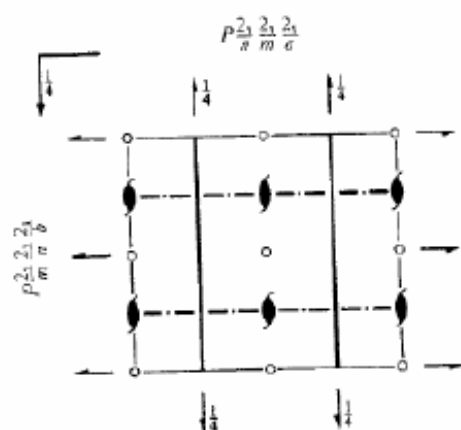
- Space group elements
- Centering
- Wyckoff positions
- Site symmetry
- Allowed Bragg peak positions

$Pnma$

No. 62

 D_{2h}^{16} $P 2_1/n 2_1/m 2_1/a$ mmm

Orthorhombic

Patterson symmetry $Pmmm$ Origin at $\bar{1}$ on 12_1 Asymmetric unit $0 \leq x \leq \frac{1}{2}; 0 \leq y \leq \frac{1}{2}; 0 \leq z \leq 1$

Symmetry operations

- | | | | | | | | |
|---------------|--------------------------|-------------------|--------------------------|---------|--------------------------|------------------------------------|-------------------|
| (1) 1 | (2) $2(0,0,\frac{1}{2})$ | $\frac{1}{2},0,z$ | (3) $2(0,\frac{1}{2},0)$ | $0,y,0$ | (4) $2(\frac{1}{2},0,0)$ | $x,\frac{1}{2},\frac{1}{2}$ | |
| (5) $\bar{1}$ | $0,0,0$ | (6) a | $x,y,\frac{1}{2}$ | (7) m | $x,\frac{1}{2},z$ | (8) $n(0,\frac{1}{2},\frac{1}{2})$ | $\frac{1}{2},y,z$ |

Generators selected (1); $t(1,0,0)$; $t(0,1,0)$; $t(0,0,1)$; (2); (3); (5)

Positions

Multiplicity,
Wyckoff letter,
Site symmetry

Coordinates

Reflection conditions

General:

8 *d* 1 (1) x, y, z (2) $\bar{x} + \frac{1}{2}, \bar{y}, z + \frac{1}{2}$ (3) $\bar{x}, y + \frac{1}{2}, \bar{z}$ (4) $x + \frac{1}{2}, \bar{y} + \frac{1}{2}, \bar{z} + \frac{1}{2}$
(5) $\bar{x}, \bar{y}, \bar{z}$ (6) $x + \frac{1}{2}, y, \bar{z} + \frac{1}{2}$ (7) $x, \bar{y} + \frac{1}{2}, z$ (8) $\bar{x} + \frac{1}{2}, y + \frac{1}{2}, z + \frac{1}{2}$

$0kl : k + l = 2n$
 $hkl : h = 2n$
 $h00 : h = 2n$
 $0k0 : k = 2n$
 $00l : l = 2n$

Special: as above, plus

4 *c* . *m* . $x, \frac{1}{4}, z$ $\bar{x} + \frac{1}{2}, \frac{3}{4}, z + \frac{1}{2}$ $\bar{x}, \frac{3}{4}, \bar{z}$ $x + \frac{1}{2}, \frac{1}{4}, \bar{z} + \frac{1}{2}$

no extra conditions

4 *b* $\bar{1}$ $0, 0, \frac{1}{2}$ $\frac{1}{2}, 0, 0$ $0, \frac{1}{2}, \frac{1}{2}$ $\frac{1}{2}, \frac{1}{2}, 0$

$hkl : h + l, k = 2n$

4 *a* $\bar{1}$ $0, 0, 0$ $\frac{1}{2}, 0, \frac{1}{2}$ $0, \frac{1}{2}, 0$ $\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$

$hkl : h + l, k = 2n$

Symmetry of special projections

Along $[001]$ $p2gm$

$\mathbf{a}' = \frac{1}{2}\mathbf{a}$ $\mathbf{b}' = \mathbf{b}$

Origin at $0, 0, z$

Along $[100]$ $c2mm$

$\mathbf{a}' = \mathbf{b}$ $\mathbf{b}' = \mathbf{c}$

Origin at $x, \frac{1}{4}, \frac{1}{4}$

Along $[010]$ $p2gg$

$\mathbf{a}' = \mathbf{c}$ $\mathbf{b}' = \mathbf{a}$

Origin at $0, y, 0$

Maximal non-isomorphic subgroups

I [2] $Pn2_1a$ ($Pna2_1$, 33) 1; 3; 6; 8
[2] $Pnm2_1$ ($Pmn2_1$, 31) 1; 2; 7; 8
[2] $P2_1ma$ ($Pmc2_1$, 26) 1; 4; 6; 7
[2] $P2_12_12_1$ (19) 1; 2; 3; 4
[2] $P112_1/a$ ($P2_1/c$, 14) 1; 2; 5; 6
[2] $P2_1/n11$ ($P2_1/c$, 14) 1; 4; 5; 8
[2] $P12_1/m1$ ($P2_1/m$, 11) 1; 3; 5; 7

IIa none

IIb none

Maximal isomorphic subgroups of lowest index

IIc [3] $Pnma$ ($\mathbf{a}' = 3\mathbf{a}$) (62); [3] $Pnma$ ($\mathbf{b}' = 3\mathbf{b}$) (62); [3] $Pnma$ ($\mathbf{c}' = 3\mathbf{c}$) (62)

Minimal non-isomorphic supergroups

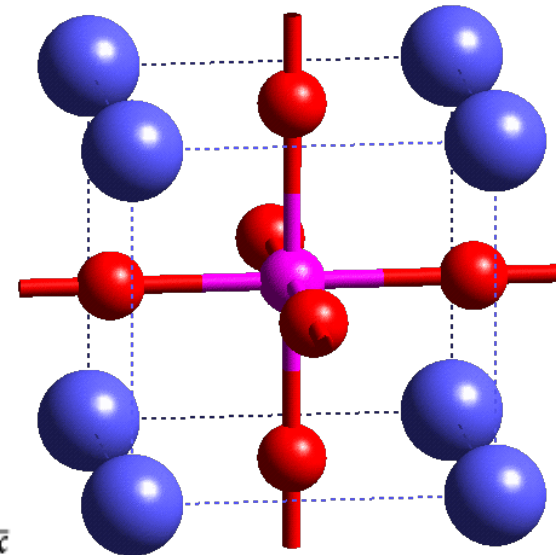
I none

II [2] $Amma$ ($Cmcm$, 63); [2] $Bbmm$ ($Cmcm$, 63); [2] $Ccme$ ($Cmce$, 64); [2] $Imma$ (74); [2] $Pcma$ ($\mathbf{b}' = \frac{1}{2}\mathbf{b}$) ($Pbam$, 55);
[2] $Pbma$ ($\mathbf{c}' = \frac{1}{2}\mathbf{c}$) ($Pbcm$, 57); [2] $Pnmm$ ($\mathbf{a}' = \frac{1}{2}\mathbf{a}$) ($Pmnm$, 59)


Example crystal structure

SrTiO_3 : Pm-3m (No. 221)

Atom	site	x	y	z
Sr	1b	0.5	0.5	0.5
Ti	1a	0	0	0
O	3d	0.5	0	0



8	g	$.3m$	x, x, x x, x, \bar{x}	\bar{x}, \bar{x}, x $\bar{x}, \bar{x}, \bar{x}$	\bar{x}, x, \bar{x} x, \bar{x}, x	x, \bar{x}, \bar{x} \bar{x}, x, x		
6	f	$4m.m$	$x, \frac{1}{2}, \frac{1}{2}$	$\bar{x}, \frac{1}{2}, \frac{1}{2}$	$\frac{1}{2}, x, \frac{1}{2}$	$\frac{1}{2}, \bar{x}, \frac{1}{2}$	$\frac{1}{2}, \frac{1}{2}, x$	$\frac{1}{2}, \frac{1}{2}, \bar{x}$
6	e	$4m.m$	$x, 0, 0$	$\bar{x}, 0, 0$	$0, x, 0$	$0, \bar{x}, 0$	$0, 0, x$	$0, 0, \bar{x}$
3	d	$4/m m.m$	$\frac{1}{2}, 0, 0$	$0, \frac{1}{2}, 0$	$0, 0, \frac{1}{2}$			
3	c	$4/m m.m$	$0, \frac{1}{2}, \frac{1}{2}$	$\frac{1}{2}, 0, \frac{1}{2}$	$\frac{1}{2}, \frac{1}{2}, 0$			
1	b	$m\bar{3}m$	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$					
1	a	$m\bar{3}m$	$0, 0, 0$					



Perovskite structure

Perovskite structure

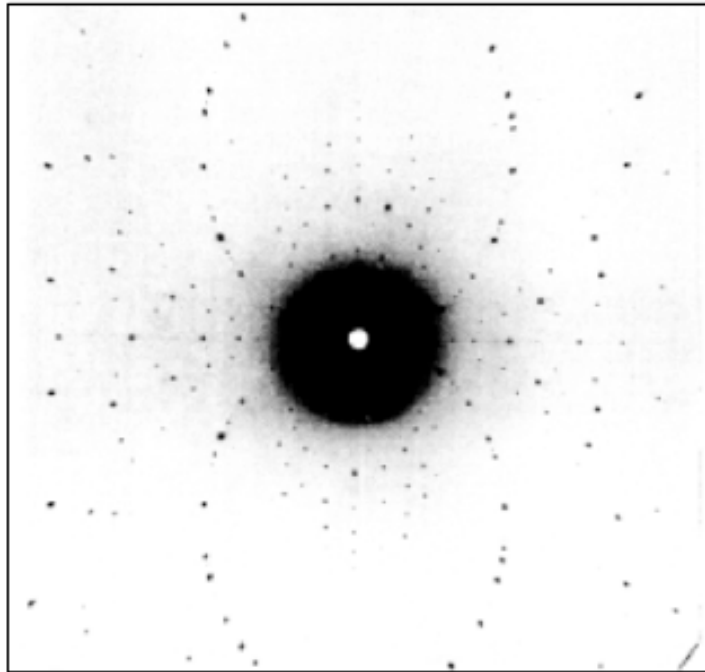
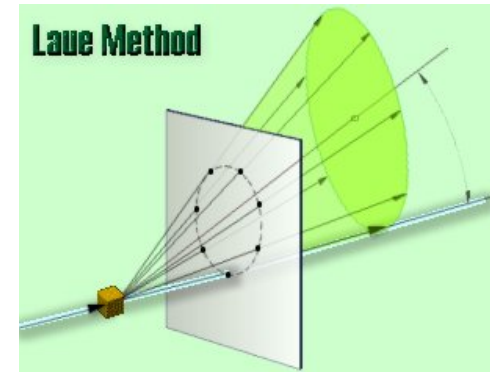
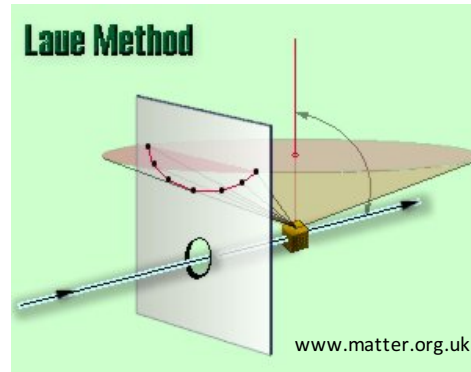
Two classes of neutron diffraction experiments

- Monochromatic beam
 - powder neutron diffraction
 - single crystal neutron diffraction
- Polychromatic beam
 - Laue diffraction
 - Time-of-flight diffraction

Laue measurements

Polychromatic beam fulfills the Bragg conditions for many reciprocal lattice vectors

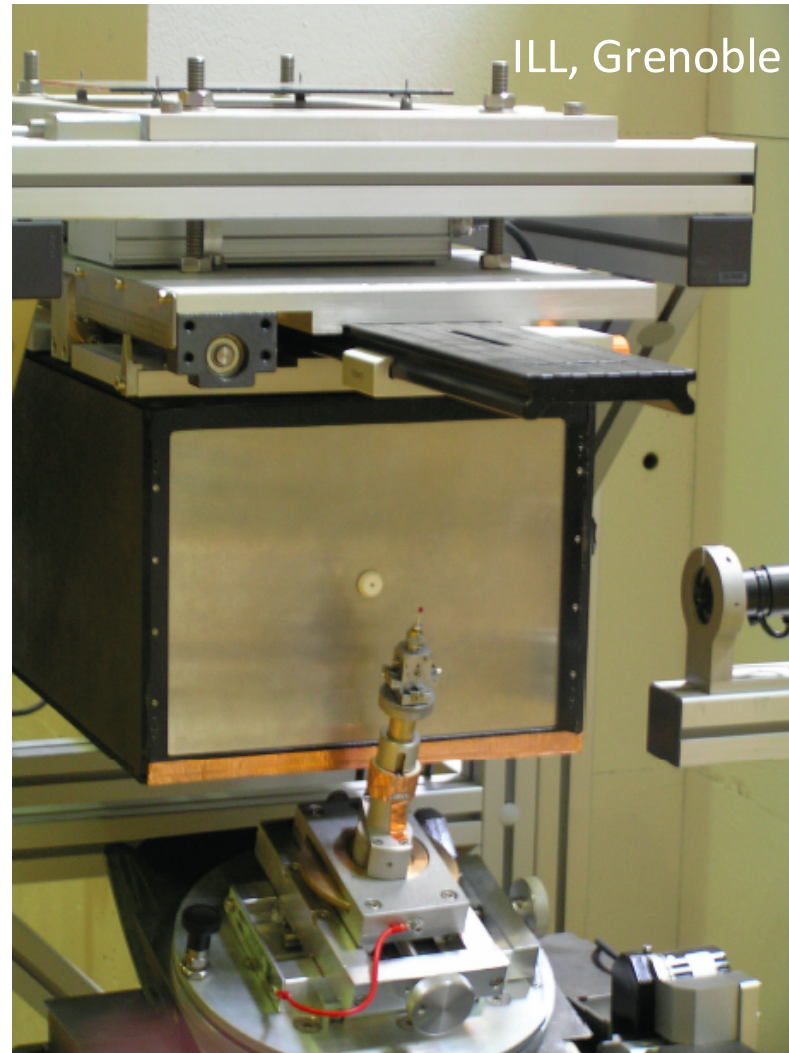
$$Q = 2k_i \sin 2\Theta$$



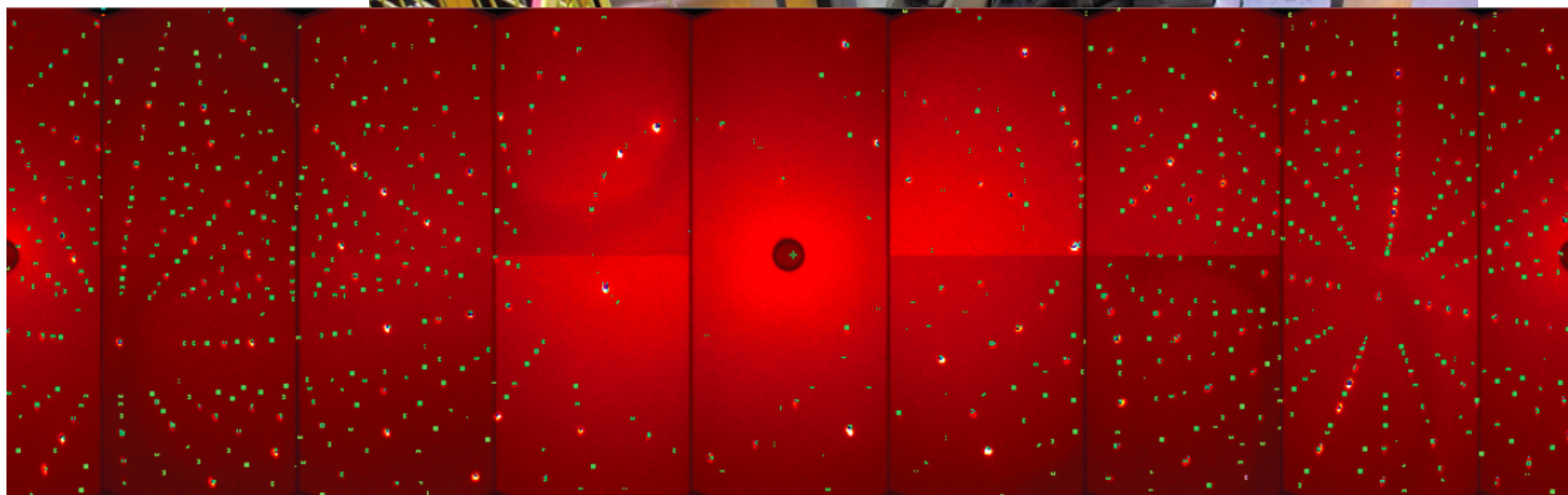
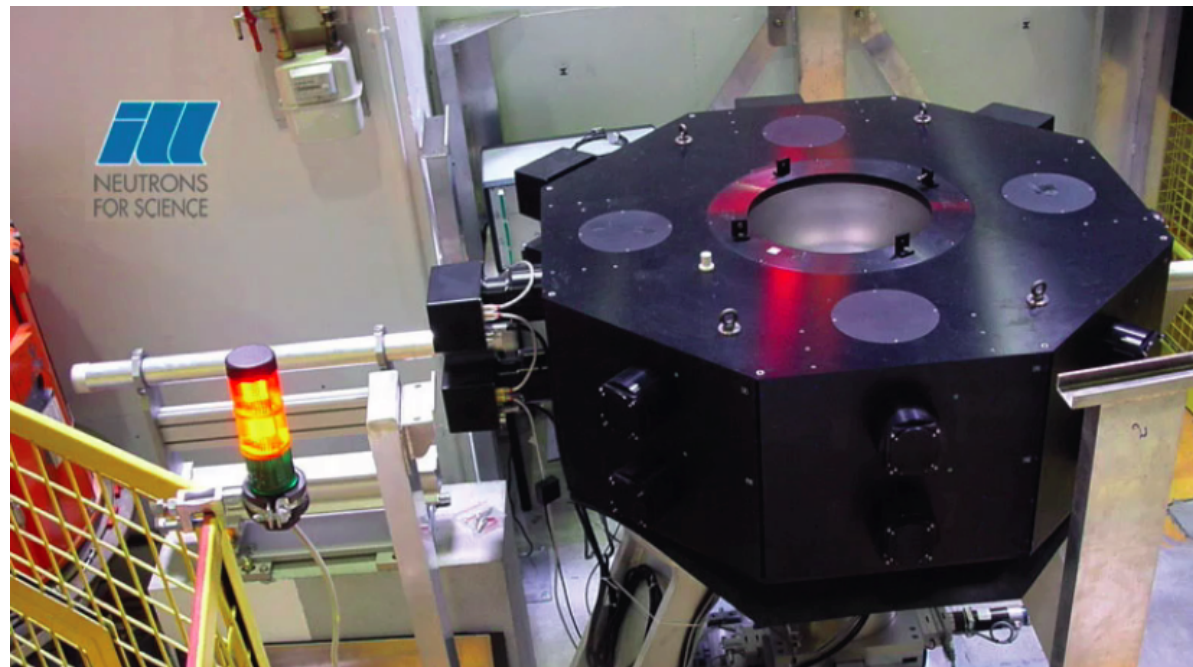
Systematic absences and symmetry of reciprocal lattice become immediately visible

Experiments are usually carried out with image plates and thus the intensity cannot be analyzed quantitatively

Example: neutron Laue instrument

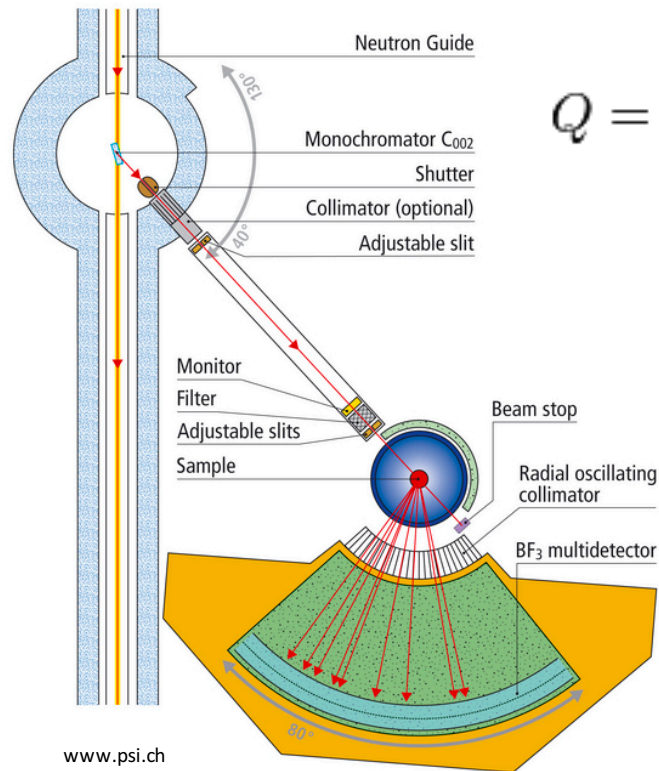


Modern neutron Laue instrument

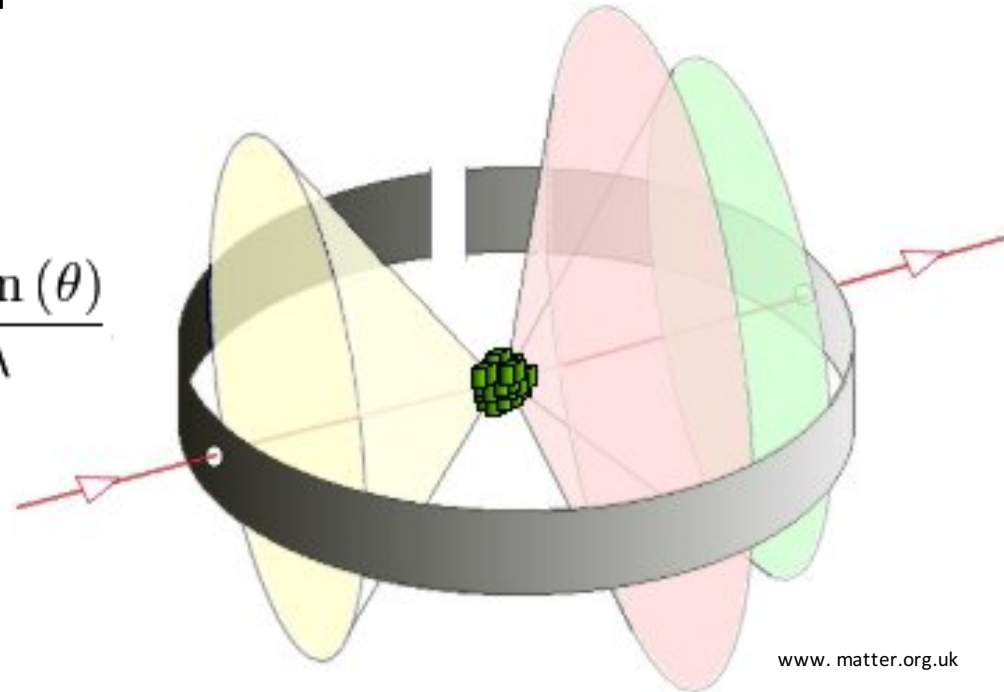


Neutron powder diffraction

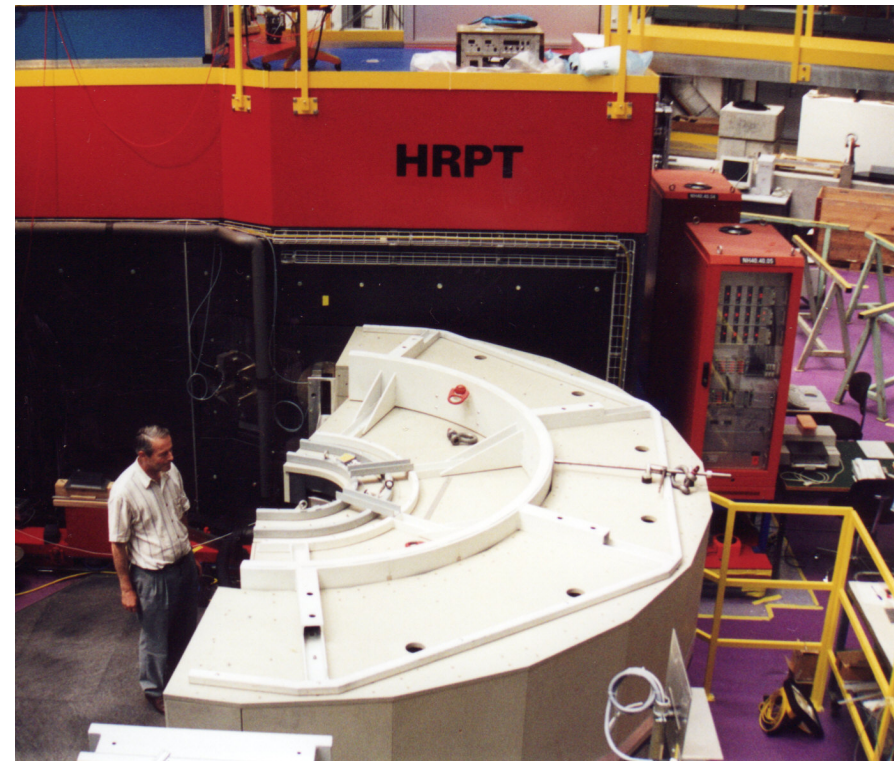
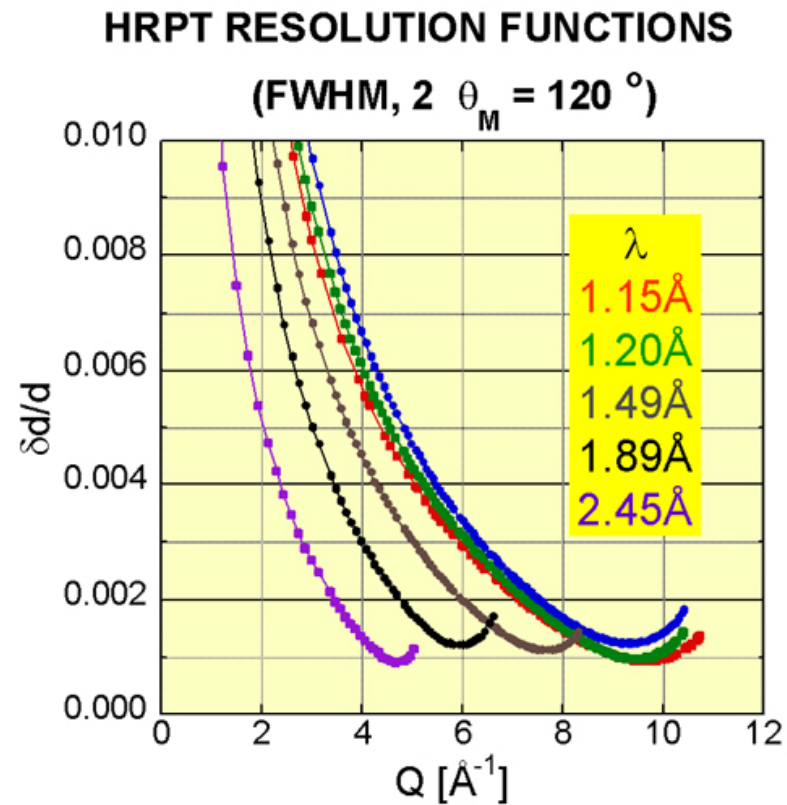
- Many materials are not available as crystals
- Polycrystalline have no orientation
- Bragg peaks are easy to find
- Data is often easier to analyze



$$Q = \frac{4\pi \sin(\theta)}{\lambda}$$



Resolution powder neutron diffraction



Phase problem

Neutron diffraction measures intensities, but not the structure factors

$$I \propto |F_N(\boldsymbol{\tau})|^2 \qquad F_N(\boldsymbol{\tau}) = \sum_d b_d e^{i\boldsymbol{\tau} \cdot \mathbf{d}}$$

Structure factors are measured without the phase

Structure cannot be simply obtained from a Fourier Transform of structure factors, because the phases are not measured

→ Fitting of models to the observed data is necessary

Powder diffraction analysis

$$I_i = a + 2b\theta_i + m_{\text{hkl}} |F_N(\tau)|^2 L(\theta_i) \exp \left(-4 \ln 2 \left(\frac{2\theta_i - 2\theta_{\text{hkl}}}{\Gamma_i} \right)^2 \right)$$

$$L(\theta) = 1/(\sin \theta \sin 2\theta)$$

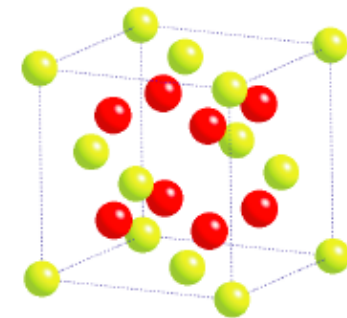
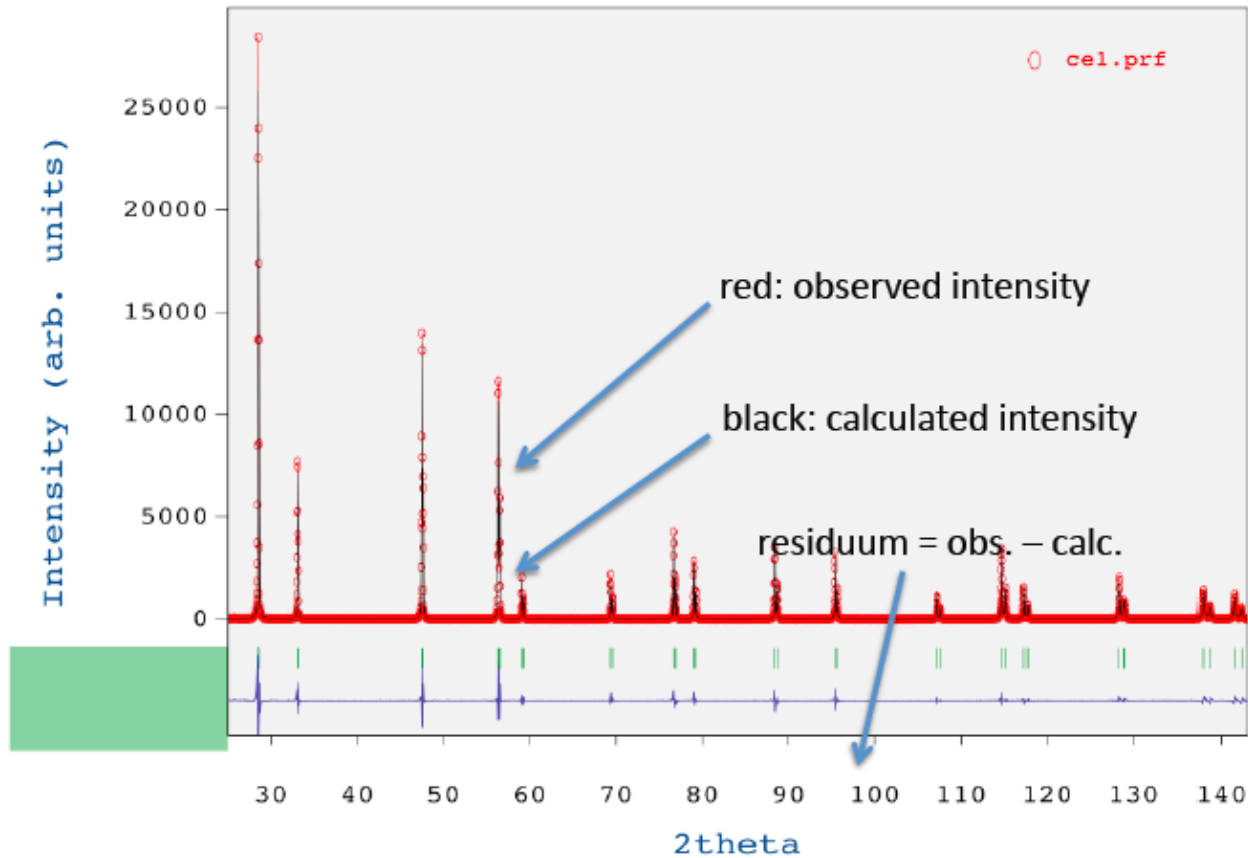
Lorentz factor arising from the Debye-Scherrer cones intersecting powder sample

$$\Gamma(2\theta) = \sqrt{U \tan^2 \theta + V \tan \theta + W}$$

Bragg peaks are broadened by resolution (finite wave-length distribution etc)

Additional information may be extracted about strain, preferred orientation etc through a peak shape analysis

Example CeO₂

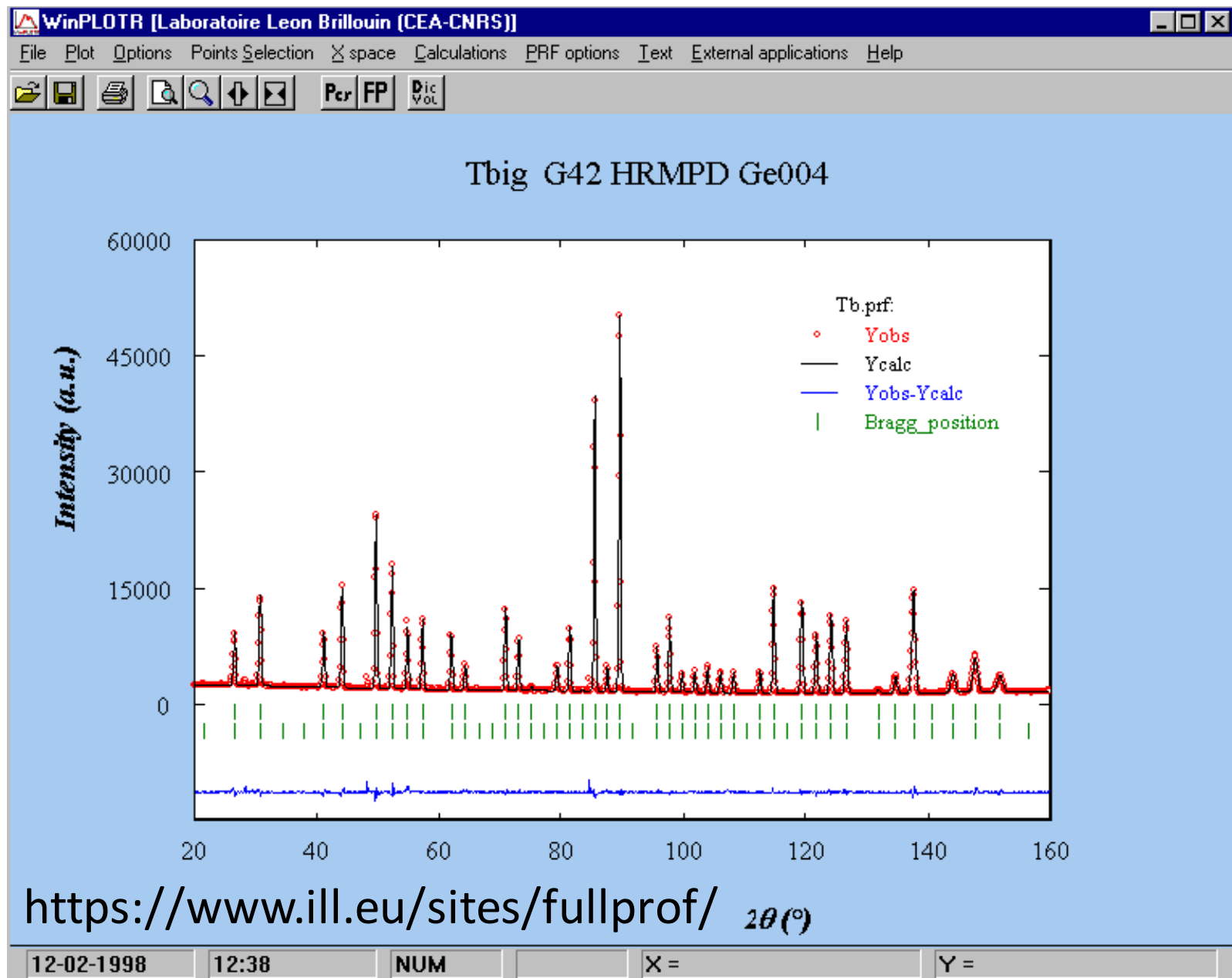


CeO₂ cubic Fm3m
Ce: 0, 0, 0
O: ¼, ¼, ¼,

Minimize

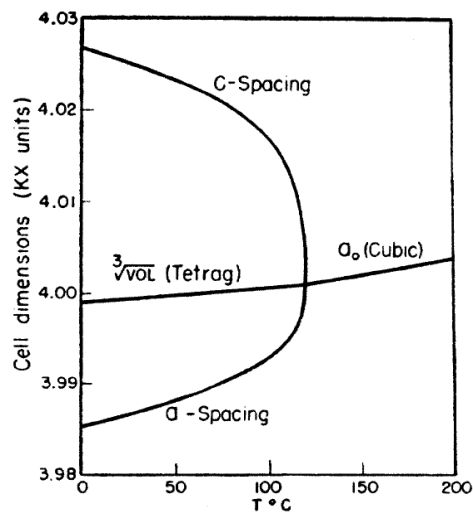
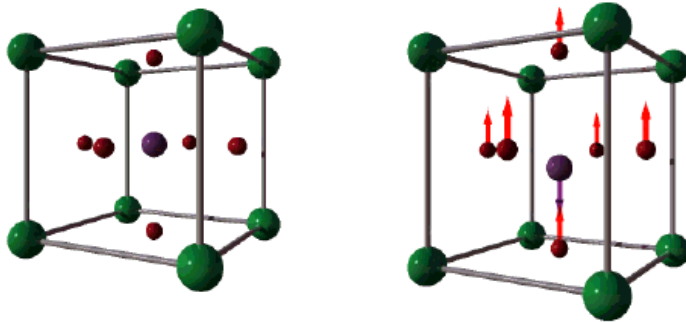
$$\chi^2 = \sum_i w_i (I_i^{\text{obs}} - I_i^{\text{cal}})^2$$

Fullprof Rietveld fitting program

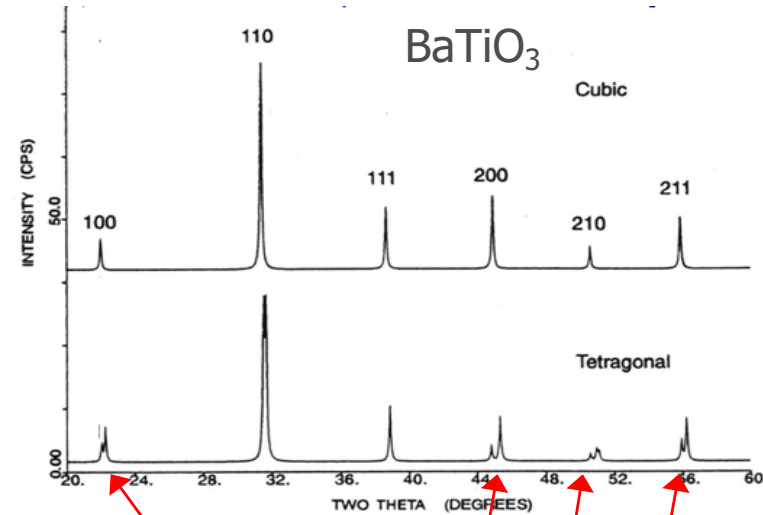


Temperature dependence of crystal structures

Onset of ferroelectricity is associated with chemical phase transition, change of lattice



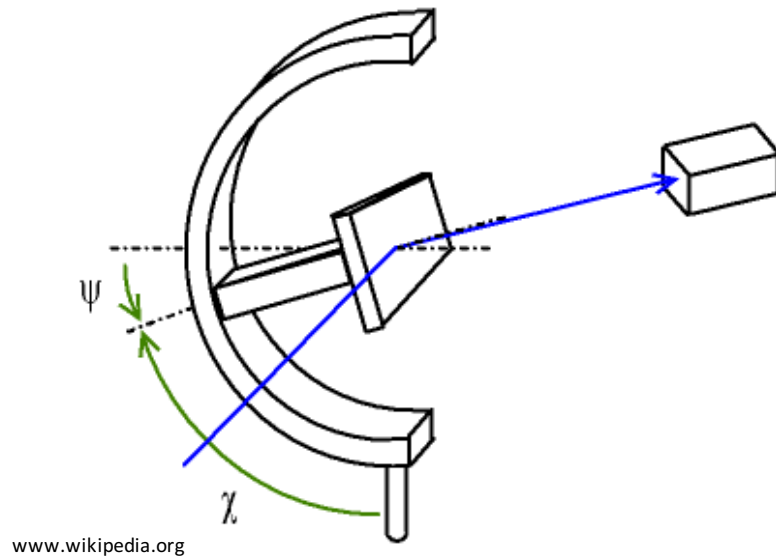
Megaw et al



Splitting of nuclear Bragg peaks

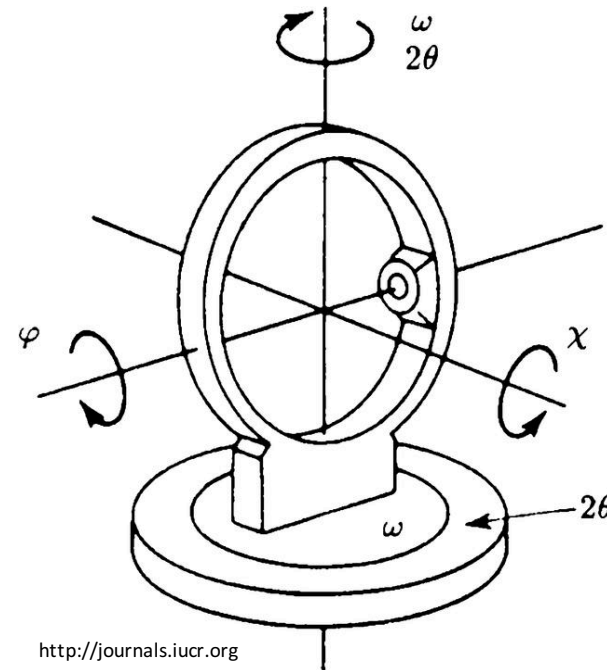
- loss of all 3-fold rotation axes
- loss of inversion symmetry
- loss of some 4-fold rotation axes

Single crystal diffraction



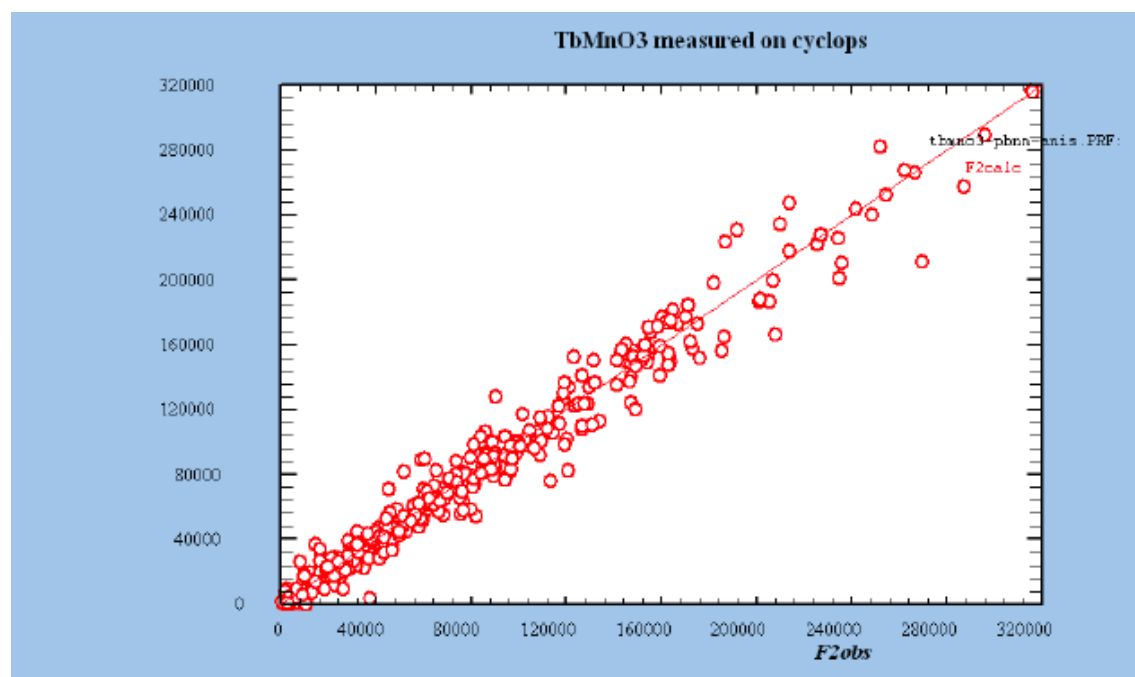
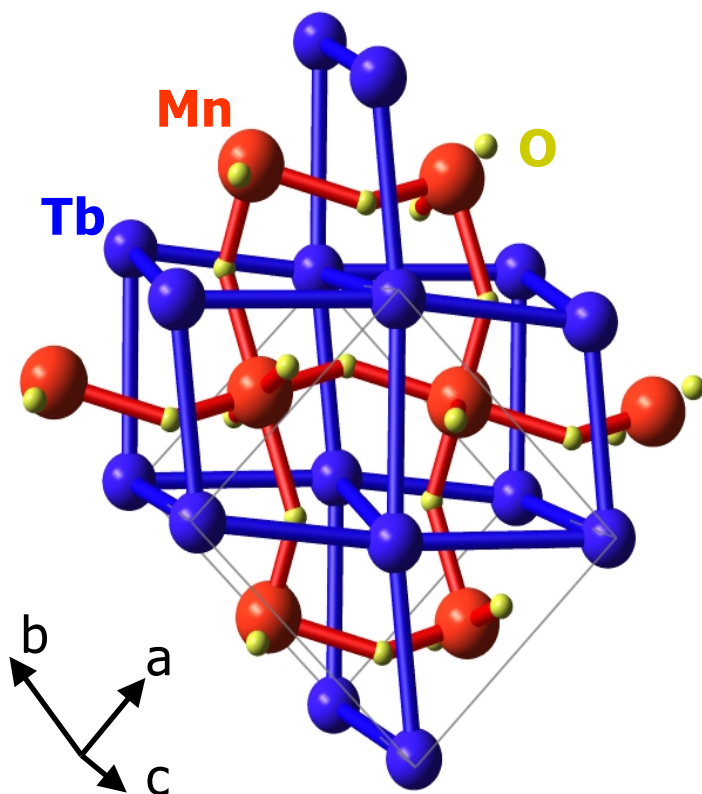
- Monochromatic beam
- Horizontal scattering plane

four angles are needed
to access all directions in
reciprocal space for
horizontal scattering plane



Lifting arm detectors allow limited access to 3D
reciprocal space only with omega and 2Theta angles

Example TbMnO₃

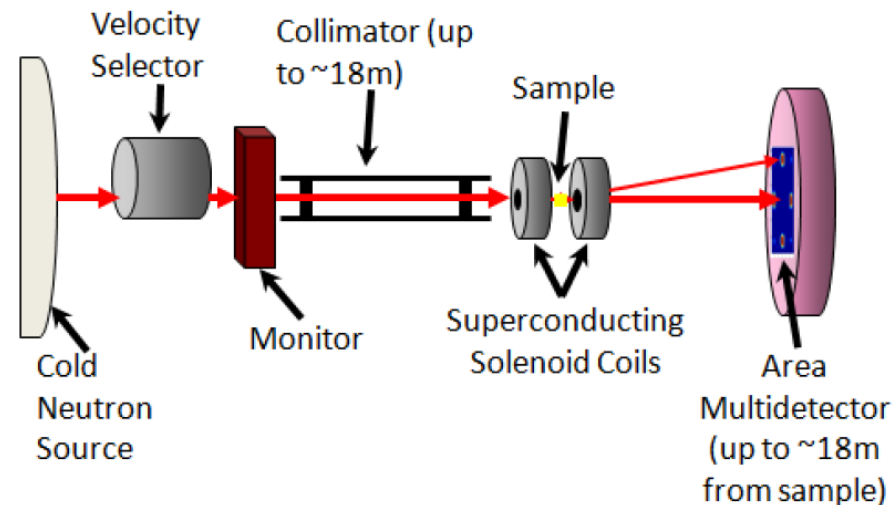


Small-angle diffraction

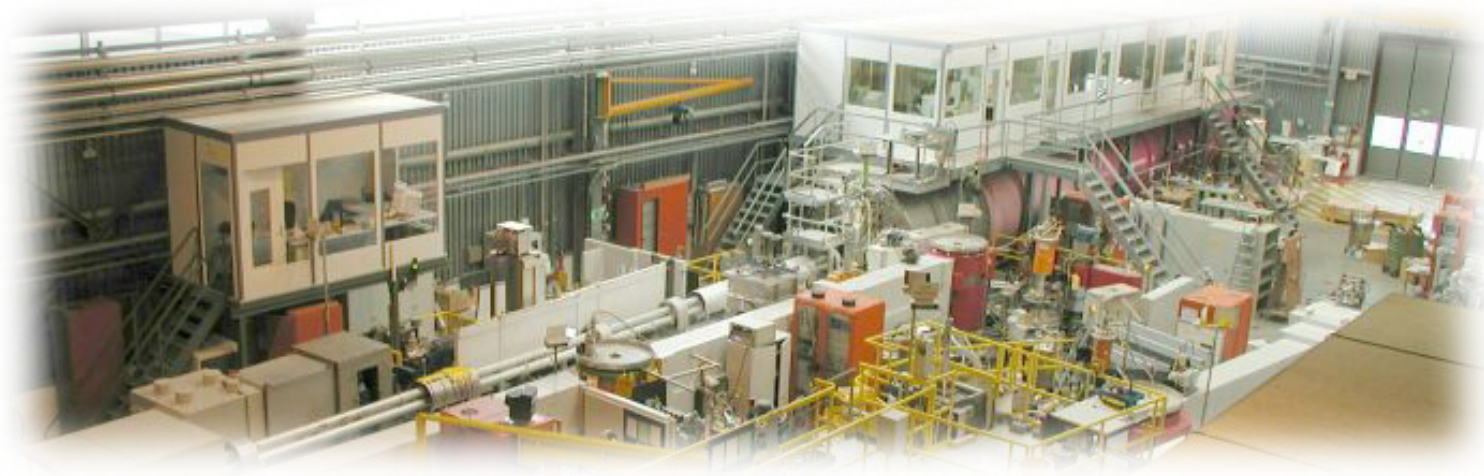
- For a unit cell of 1 nm and neutron wave-length of 5 Angstrom, lowest angle peaks are at $2\Theta = 29^\circ$

$$(n)\lambda = 2d\sin(\theta)$$

- If a unit cell is large, then 2 Theta becomes small
for $d=20\text{nm} \rightarrow 2\Theta \sim 1^\circ$
- Such low-angle scattering experiment are performed on small-angle neutron scattering (SANS) instruments



SANS instrument



SANS-I beamline at Paul Scherrer Institute, Villigen (Switzerland).

Measurement of structure of a flux lattice in a superconductor

