

Testing fundamental interactions with few electron atoms and molecules

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Motivation and challenges

Motivations

- *the interface between the strong and electromagnetic interactions, can be probed very accurately in atomic physics*
- how come r_p from (electronic) H differs by 4% from that of μH ?
- violation of lepton universality of SM ?
- incorrect Ry constant ?
 - muonium 1S-2S + HFS, muon life time $2.2 \mu\text{s}$
 - $\text{He}^+(1\text{S}-2\text{S})$, T. Udem (Garching), K. Eikema (Amsterdam)
 - highly excited states of heavy hydrogen-like ions (NIST)

Few electron atomic and molecular systems

The general purpose of our project is to bring the high accuracy achieved for hydrogenic levels to few-electron atomic and molecular systems

- for the determination of the nuclear charge radii from experimental transition frequencies
- to search for any discrepancies with spectroscopic measurements and uncover unknown forces at the atomic scale
- to carry out low-energy tests and of various extensions of the Standard Model of fundamental interactions
- determination of fundamental constants Ry , α

$$f(1S - 2S, H) = 2\,466\,061\,413\,187\,035(10)\text{Hz} \quad (\sim 10^{-16})$$

lack of second sharp transition in H

- H_2 dissociation energy F. Merkt (ETH) + W. Ubachs (Amsterdam)

The nuclear finite size effect

- $\delta_{\text{fs}} E = \frac{2\pi Z\alpha}{3} \phi^2(0) \langle r_{\text{ch}}^2 \rangle = C \langle r_{\text{ch}}^2 \rangle$
- this formula is universal, valid for an arbitrary atomic system
- higher order $O(Z\alpha r_{\text{ch}}/\lambda)$, small for electronic atoms
- determination of mean square nuclear charge radius from:

$$\langle r_{\text{ch}}^2 \rangle = \frac{E_{\text{exp}} - E_{\text{the}}}{C}$$

α charge radius from He $2^3S - 2^3P$

- $E(2^3S - 2^3P, {}^4\text{He})_{\text{centroid}} = 276\,736\,495\,649.5(2.1)$ kHz, Florence, 2004
- finite size effect: $E_{\text{fs}} = 3\,427$ kHz
- since E_{fs} is proportional to r^2

$$\frac{\Delta r}{r} = \frac{1}{2} \frac{\delta E_{\text{fs}}}{E_{\text{fs}}} \approx \frac{1}{2} \frac{10}{3\,427} = 1.5 \cdot 10^{-3}$$

- electron scattering gives $r_{\text{He}} = 1.681(4)$ fm, what corresponds to about $2.5 \cdot 10^{-3}$ relative accuracy
- ~ 10 kHz accuracy requires calculation of $m\alpha^7$ correction

r_P from dissociation energy of H_2

- $D_0(H_2) = 36118.06962(37) \text{ cm}^{-1}$, J. Liu et al, 2009
- at the $6 \cdot 10^{-7} \text{ cm}^{-1}$ accuracy for H_2 levels
- the proton charge radius which contributes $3.1 \cdot 10^{-5} \text{ cm}^{-1}$ to the dissociation energy can be determined to 1.0% accuracy (discrepancy is at 4%)
- requires calculation of $m\alpha^7$ and the so called nonadiabatic corrections.

Hydrogen theory

- Dirac equation (with the infinite nuclear mass)
- one loop self-energy and vacuum polarization

$$E = \frac{\alpha}{\pi} (Z\alpha)^4 \left[A_{40} + (Z\alpha) A_{50} + (Z\alpha)^2 (A_{62} \ln^2(Z\alpha)^{-2} + A_{61} \ln(Z\alpha)^{-2} + A_{60}) + \dots \right]$$

or directly numerically, known very accurately

- two-loop correction

$$E = \left(\frac{\alpha}{\pi} \right)^2 (Z\alpha)^4 \left[B_{40} + (Z\alpha) B_{50} + (Z\alpha)^2 (B_{63} \ln^3(Z\alpha)^{-2} + B_{62} \ln^2(Z\alpha)^{-2} + B_{61} \ln(Z\alpha)^{-2} + B_{60}) + \dots \right]$$

B_{60} is not very accurate, main source of the uncertainty, a few kHz for 1S

- pure (Shabaev formula) and radiative recoil

NRQED approach to few electron systems

Atomic energy levels are expanded in powers of **the fine structure constant α**

$$E(\alpha) = E^{(2)} + E^{(4)} + E^{(5)} + E^{(6)} + E^{(7)} + \dots, \quad E^{(n)} \sim m \alpha^n$$

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$E^{(2)}$ is a **nonrelativistic energy** corresponding to the Hamiltonian

$$H^{(2)} = \sum_a \frac{\vec{p}_a^2}{2m} - \frac{Z\alpha}{r_a} + \sum_{a>b} \frac{\alpha}{r_{ab}}$$

- All expansion terms are expressed in terms of expectation values of some effective Hamiltonian with the nonrelativistic wave function

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Leading relativistic correction

$$E^{(4)} = \langle H^{(4)} \rangle$$

$$\begin{aligned}
 H^{(4)} = & \sum_a \left\{ -\frac{\vec{p}_a^4}{8m^3} + \frac{\pi Z \alpha}{2m^2} \delta^3(r_a) + \frac{Z \alpha}{4m^2} \vec{\sigma}_a \cdot \frac{\vec{r}_a}{r_a^3} \times \vec{p}_a \right\} \\
 & + \sum_{a>b} \sum_b \left\{ -\frac{\pi \alpha}{m^2} \delta^3(r_{ab}) - \frac{\alpha}{2m^2} p_a^j \left(\frac{\delta^{ij}}{r_{ab}} + \frac{r_{ab}^i r_{ab}^j}{r_{ab}^3} \right) p_b^j \right. \\
 & - \frac{2\pi \alpha}{3m^2} \vec{\sigma}_a \cdot \vec{\sigma}_b \delta^3(r_{ab}) + \frac{\alpha}{4m^2} \frac{\sigma_a^i \sigma_b^j}{r_{ab}^3} \left(\delta^{ij} - 3 \frac{r_{ab}^i r_{ab}^j}{r_{ab}^2} \right) + \frac{\alpha}{4m^2 r_{ab}^3} \\
 & \left. \times \left[2 \left(\vec{\sigma}_a \cdot \vec{r}_{ab} \times \vec{p}_b - \vec{\sigma}_b \cdot \vec{r}_{ab} \times \vec{p}_a \right) + \left(\vec{\sigma}_b \cdot \vec{r}_{ab} \times \vec{p}_b - \vec{\sigma}_a \cdot \vec{r}_{ab} \times \vec{p}_a \right) \right] \right\}
 \end{aligned}$$

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Leading QED correction

$$\begin{aligned} E^{(5)} = & \left[\frac{164}{15} + \frac{14}{3} \ln \alpha \right] \frac{\alpha^2}{m^2} \langle \delta^3(r_{12}) \rangle \\ & + \left[\frac{19}{30} + \ln(Z\alpha)^{-2} \right] \frac{4\alpha^2 Z}{3m^2} \langle \delta^3(r_1) + \delta^3(r_2) \rangle \\ & - \frac{14}{3} m \alpha^5 \left\langle \frac{1}{4\pi} P \left(\frac{1}{(m\alpha r_{12})^3} \right) \right\rangle \\ & - \frac{2\alpha}{3\pi m^2} \left\langle \sum_a \vec{p}_a (H_0 - E_0) \ln \left[\frac{2(H_0 - E_0)}{(Z\alpha)^2 m} \right] \sum_b \vec{p}_b \right\rangle \end{aligned}$$

NRQED approach to few electron systems

Atomic energy levels are expanded in powers of **the fine structure constant α**

$$E(\alpha) = E^{(2)} + E^{(4)} + E^{(5)} + E^{(6)} + E^{(7)} + \dots, \quad E^{(n)} \sim m \alpha^n$$

Higher order effects $m\alpha^6, m\alpha^7 \dots$

- $E^{(6)} = \langle H^{(6)} \rangle + \langle H^{(4)} \frac{1}{(E_0 - H_0)^T} H^{(4)} \rangle$
- cancellation of singularities between the first and the second order matrix elements \rightarrow difficult in numerical calculations
- $E^{(7)}$ known only for hydrogenic systems (+ He fs) \rightarrow challenging task for few electron atoms

$2^3S - 2^3P$ transition in ^4He in MHz

	$(m/M)^0$	$(m/M)^1$	$(m/M)^2$	Sum
α^2	-276 775 637.536	102 903.459	-4.781	-276 672 738.857
α^4	-69 066.189	-6.769	-0.003	-69 072.961
α^5	5 234.163	-0.186	—	5 233.978
α^6	87.067	-0.029	—	87.039
α^7	-8.0 (1.0)	—	—	-8.0(1.0)
FNS	3.427	—	—	3.427
NPOL	-0.002	—	—	-0.002
Theory				-276 736 495.41 (1.00)
Exp.				-276 736 495.649 (2)

$^3\text{He} - ^4\text{He}$ isotope shift of $2^3S - 2^3P$ centroid energies in kHz

	$(m/M)^1$	$(m/M)^2$	$(m/M)^3$	Sum
α^2	33 673 018.7	-3 640.6	0.4	33 669 378.5
α^4	-2 214.9	-2.4	—	-2 217.3
α^5	-60.7	—	—	-60.7
α^6	-9.4	—	—	-9.4
α^7	0.0 (0.9)	—	—	0.0 (0.9)
NPOL	-1.1	—	—	-1.1
EMIX	—	54.6	—	54.6
Theory				33 667 149.3(0.9)

$^3\text{He} - ^4\text{He}$ charge radii difference

$$\delta r^2(\text{Florence 2012}, 2^3P - 2^3S) = 1.069 (3) \text{ fm}^2,$$

$$\delta r^2(\text{Shiner 1995}, 2^3P - 2^3S) = 1.061 (3) \text{ fm}^2,$$

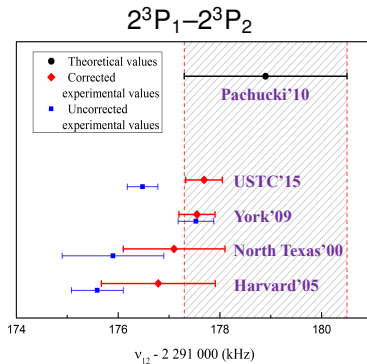
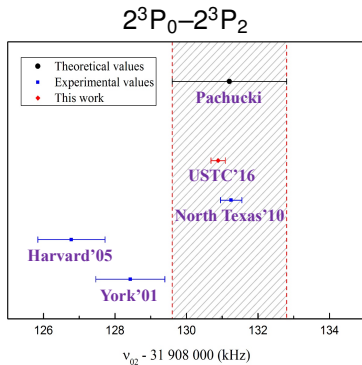
$$\delta r^2(\text{Amsterdam 2011}, 2^1S - 2^3S) = 1.028 (11) \text{ fm}^2$$

3.7 σ discrepancy

He fine structure

Term	ν_{01}	ν_{12}	Ref.
$m\alpha^4(+m/M)$	29 563 765.45	2 320 241.43	Drake (2002)
	29 563 765.23	2 320 241.42	
$m\alpha^5(+m/M)$	54 704.04	-22 544.00	Drake (2002)
	54 704.04	-22 545.01	
$m\alpha^6$	-1 607.52(2)	-6 506.43	Drake (2002)
	-1 607.61(4)	-6 506.45(7)	
$m\alpha^6 m/M$	-9.96	9.15	Drake (2002)
	-10.37(5)	9.80(11)	
$m\alpha^7 \log(Z\alpha)$	81.43	-5.87	Drake (2002)
	81.42	-5.87	
$m\alpha^7, \text{nlog}$	18.86	-14.38	
$m\alpha^8$	± 1.7	± 1.7	
Total theory	$29\,616\,952.29 \pm 1.7$	$2\,291\,178.91 \pm 1.7$	

Comparison with experiments: Yu Sun PSAS'16



Sensitive test of hypothetical additional long range interactions

Hydrogen molecule: comparison with experiment

H₂ has not been considered so far for high precision test of fundamental interactions

	D_0	$\nu = 0 \rightarrow 1$	$J = 0 \rightarrow 1$
$\alpha^2 m$	36 118.797 746 12(5)	4 161.164 070 3(1)	118.485 260 46(3)
$\alpha^4 m$	-0.531 8(3) ^a	0.023 41(1) ^c	0.002 580(1)
$\alpha^5 m$	-0.194 8(2) ^b	-0.021 29(2) ^c	-0.001 022(1)
$\alpha^6 m$	-0.002 065(6)	-0.000 192 3(6)	-0.000 008 9(1)
$\alpha^7 m$	0.000 118(59)	0.000 012 0(60)	0.000 000 6(3)
$E_{r_p^2}$	-0.000 031	-0.000 003 2	-0.000 000 2
Theory	36 118.069 1(6)	4 161.166 01(4)	118.486 810(4)
Exp	36 118.069 62(37)	4 161.166 32(18)	118.486 84(10)

D_0 - dissociation energy in cm⁻¹

Dissociation energy – comparison with experiment

	D_0/cm^{-1}	
	H ₂	D ₂
Experiment (1993) ¹	36 118.06(4)	36 748.32(7)
Experiment (2004) ²	36 118.062(10)	36 748.343(10)
Experiment (2009/10) ^{3,4}	36 118.069 62(37)	36 748.362 86(68)
Theory (2016), (2009) ⁵	36 118.069 1(6)	36 748.363 4(9)
Difference	0.000 5(7)	0.000 5(11)

¹E. E. Eyler, N. Melikechi, *Phys. Rev. A* **48**, R18 (1993);

²Y. Zang *et al.*, *Phys. Rev. Lett.* **92**, 203003 (2004);

³Liu, Salumbides, Hollenstein, Koelemeij, Eikema, Ubachs, Merkt, *JCP* **130**, 174306 (2009)

⁴Liu, Sprecher, Jungen, Ubachs, Merkt, *JCP* **132**, 154301 (2010);

⁵Piszczatowski, Lach, Przybytek, Komasa, Pachucki, Jeziorski, *JCTC* **5**, 3039 (2009)

Conclusions

The high accuracy achieved for few-electron atomic and molecular systems allows

- determination of the nuclear charge radii from experimental transition frequencies
- searching for any discrepancies with spectroscopic measurements and uncover unknown forces at the atomic scale
- low-energy tests of the Standard Model of fundamental interactions and of its various extensions

Collaborators

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