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Extended defects modelling of functional oxides *the intriguing case of UO_2*

Empa Postdocs-II & PSI-FELLOW II-3i RETREAT 2018

Functional Oxides in Emerging Technologies



**Li-ion Battery
Cathodes**

**All-Oxide
Photovoltaics
Light
absorbers
Charge
conductors**

**Fuel Cells
Non precious
metal electro
catalysts**

**Nuclear Fuels
safety, UO_2**

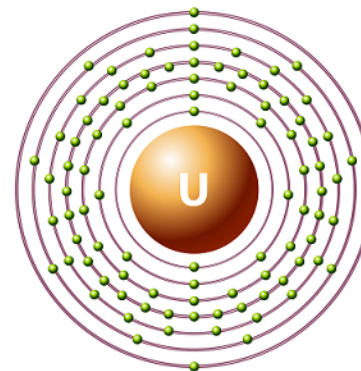
Unifying Motif: complex electronic structure enables functionality

Practical Utilization: how MATERIALS respond to stress?

Functional Oxides – Complex Electronic Structure, UO₂ nuclear fuel

Periodic Table of the Elements

1																	18														
1 H 1.008																	2 He 4.003														
3 Li 6.941	4 Be 9.012											13 B 10.81	14 C 12.01	15 N 14.01	16 O 16.00	17 F 19.00	18 Ne 20.18														
11 Na 22.99	12 Mg 24.31											13 Al 26.98	14 Si 28.09	15 P 30.97	16 S 32.07	17 Cl 35.45	18 Ar 39.95														
19 K 39.10	20 Ca 40.08	21 Sc 44.96	22 Ti 47.88	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.69	29 Cu 63.55	30 Zn 65.39	31 Ga 69.72	32 Ge 72.61	33 As 74.92	34 Se 78.96	35 Br 79.90	36 Kr 83.80														
37 Rb 85.49	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb 92.91	42 Mo 95.94	43 Tc 98.91	44 Ru 101.1	45 Rh 102.9	46 Pd 106.4	47 Ag 107.9	48 Cd 112.4	49 In 114.8	50 Sn 118.7	51 Sb 121.8	52 Te 127.6	53 I 126.9	54 Xe 131.3														
55 Cs 132.9	56 Ba 137.3	57 La 138.9	58 Ce 140.1	59 Pr 140.9	60 Nd 144.2	61 Pm 144.9	62 Sm 150.4	63 Eu 152.0	64 Gd 157.3	65 Tb 158.9	66 Dy 162.5	67 Ho 164.9	68 Er 167.3	69 Tm 168.9	70 Yb 173.0	71 Lu 174.9	72 Hf 178.5	73 Ta 180.9	74 W 183.8	75 Re 186.2	76 Os 190.2	77 Ir 192.2	78 Pt 195.1	79 Au 197.0	80 Hg 200.6	81 Tl 204.4	82 Pb 207.2	83 Bi 209.0	84 Po 210.0	85 At 210.0	86 Rn 222.0
87 Fr 223.0	88 Ra 226.0	89 Ac 227.0	90 Th 232.0	91 Pa 231.0	92 U 238.0	93 Np 237.0	94 Pu 244.1	95 Am 243.1	96 Cm 247.1	97 Bk 247.1	98 Cf 251.1	99 Es 252.1	100 Fm 257.1	101 Mt 268	102 Uun 269	103 Uuu 272	104 Uub 277	105 Uut 289	106 Uuq 289	107 Uup 289	108 Uuh 289	109 Uus 293	110 Uuo 293								



92

U

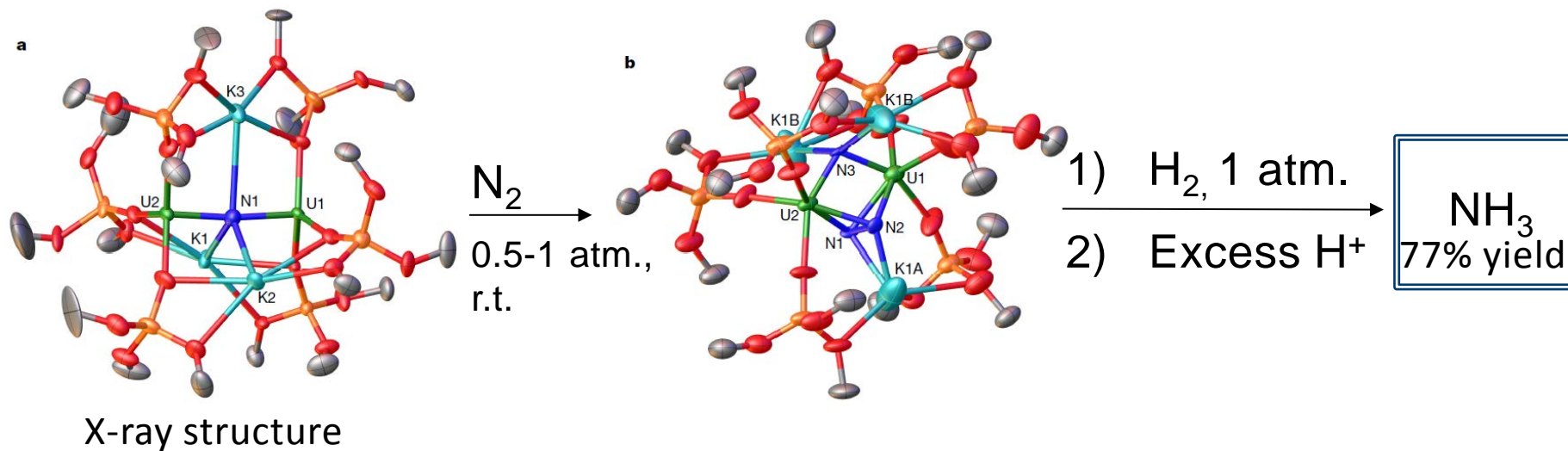
Uranium

238.02891

Atomic mass: 238.02
 Electron configuration: 2, 8, 18, 32, 21, 9, 2

Versatility of Uranium based materials *a catalyst for N₂ Functionalization*

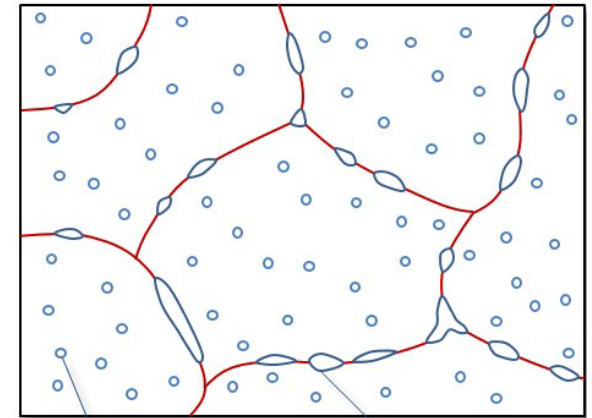
- Industrial Haber–Bosch process: $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
450°C, 300 bar.
- Biological N₂ fixation, $\text{N}_2 + 8\text{H}^+ + 8\text{e}^- \rightarrow 2\text{NH}_3 + \text{H}_2$



Nuclear Fuel UO_2

Mechanical Properties in the Plastic Regime

- Irradiation:
compositional changes
temperature jumps
- U fission process:
noble gases, Xe, Kr, release
insoluble in the fuel matrix
precipitate within the fuel body.
- Extended defects:
dislocations/grain boundaries can pin
and immobilize gas bubbles < 1 micron.



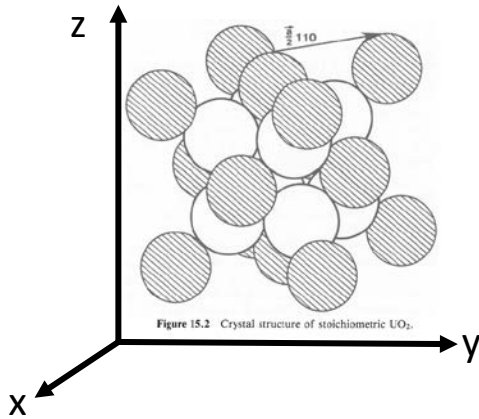
Intragranular
gas bubbles

Intergranular
gas bubbles

Structure - Property relationship of Dislocations in UO_2

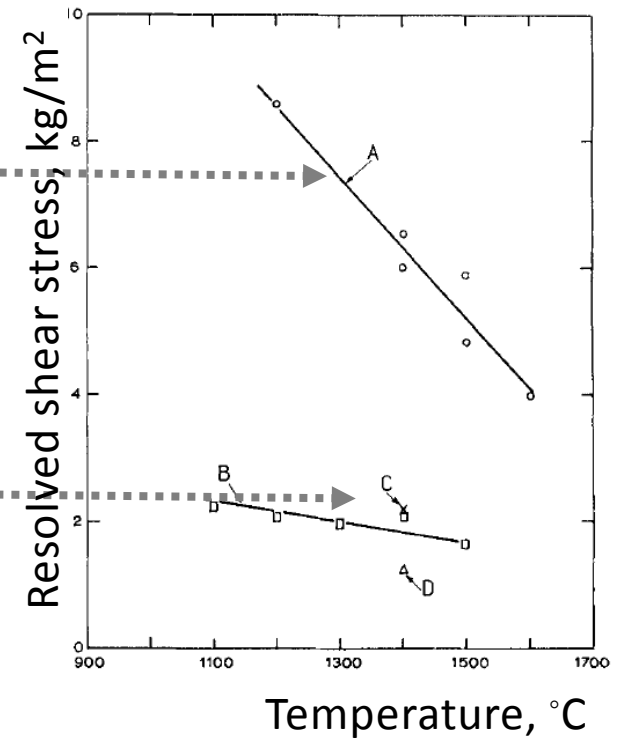
I. Experimental characterization of slip systems in UO_2 compression tests

Stoichiometric UO_2 slip systems



A difficult slip system:
 $\{110\}\langle 110\rangle$ (A)

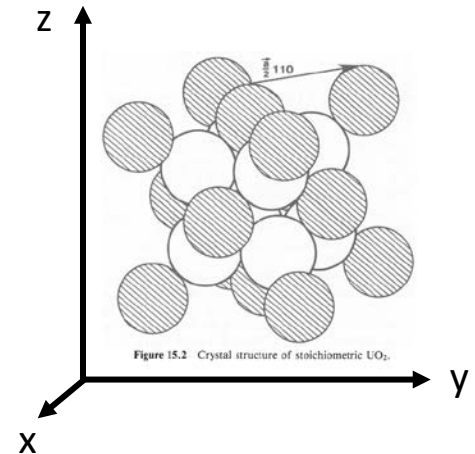
An easy slip system:
 $\{001\}\langle 110\rangle$ (C)



II. Experimental characterization of slip systems in UO_2 *compression tests*

Stoichiometric UO_2 (U/O=2.0001) at 873K
and 523K:

A **primary** slip system: $\{111\}\langle 1-10 \rangle^*$
with presence of $\{001\}\langle 1-10 \rangle$



The reason for the dominance of $\{111\}$ slip for $T < 873 \text{ K}$ is not well understood. It was demon-

Why this behavior should be observed in UO_2 , but not in CaF_2 , is not clear; we speculate that differences in details of the bonding between anions and cations (the amount of ionic vs covalent character) may be responsible.

*50% difference in yield stresses w.r.t previous experimental studies.

The Goal and the Tasks

Understand how fission product gases interact with dislocations



Understand structure property relationship of dislocation



Develop computational tools for modelling dislocations
of materials with complex electronic structure

Empirical Potentials - Force fields

$$U(\vec{R}) = \underbrace{\sum_{\text{bonds}} k_i^{\text{bond}} (r_i - r_0)^2}_{U_{\text{bond}}} + \underbrace{\sum_{\text{angles}} k_i^{\text{angle}} (\theta_i - \theta_0)^2}_{U_{\text{angle}}} + \dots$$

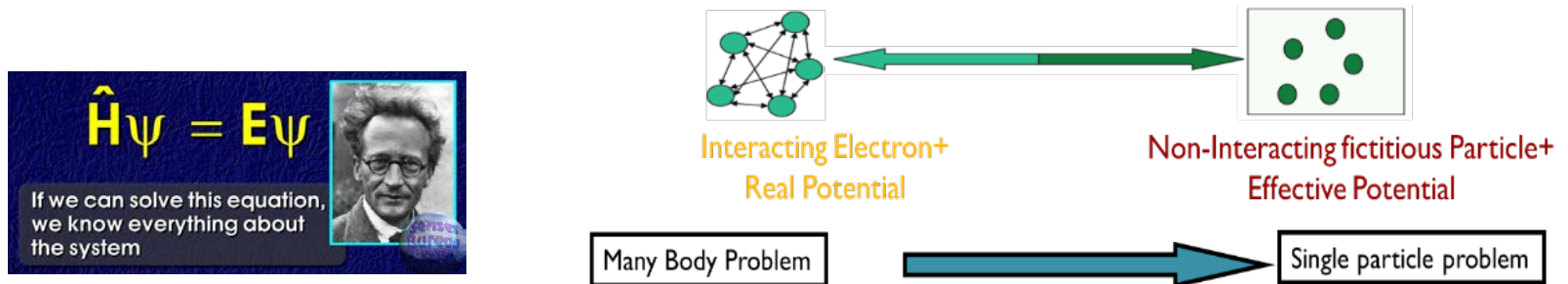
Cheap computationally: Systems of 10,000-100,000 atoms.

Neglect electronic structure: Covalence in the U-O bond and change in the oxidation state.

First principles methods: DFT+U

CPU intensive: 600-3000 atoms

Detailed electronic structure: bond formation, possible change in the oxidation state during dislocations.



f-electrons in a standard DFT:



- **dispersed among all f orbitals**

f-electrons via

***f* orbital *O*ccupation *P*attern, *FOP*:**

- restoring the band gap of UO_2
- magnetic structure

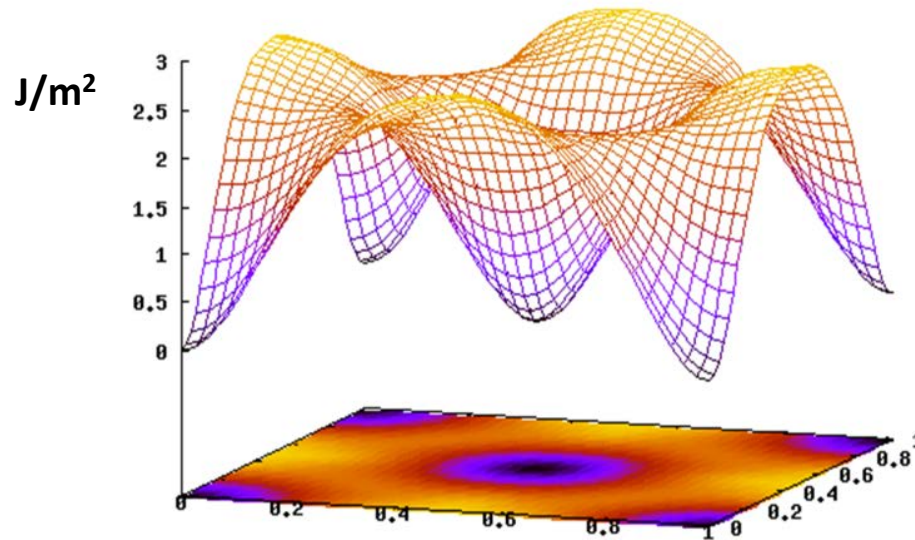


- **f orbitals' localized electrons**

Computational studies of dislocations in UO_2

general computational scheme

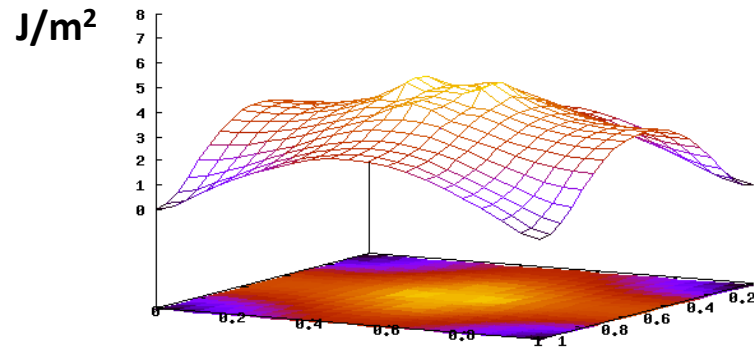
Plotting energy of crystal as function of displacement



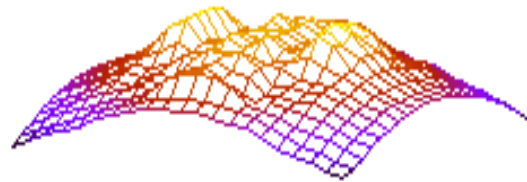
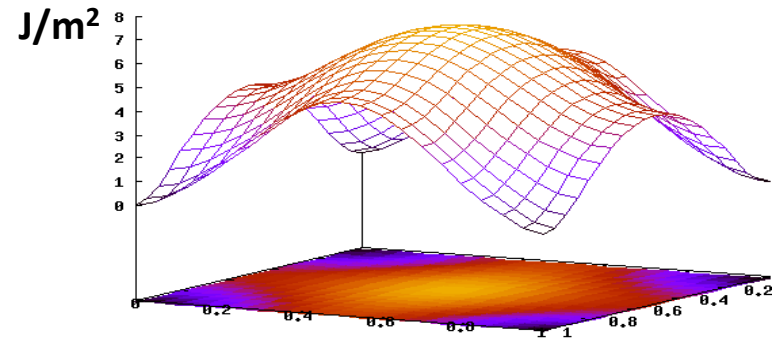
γ -surface, Generalised Stacking Fault Energies

{110}

PBE+U



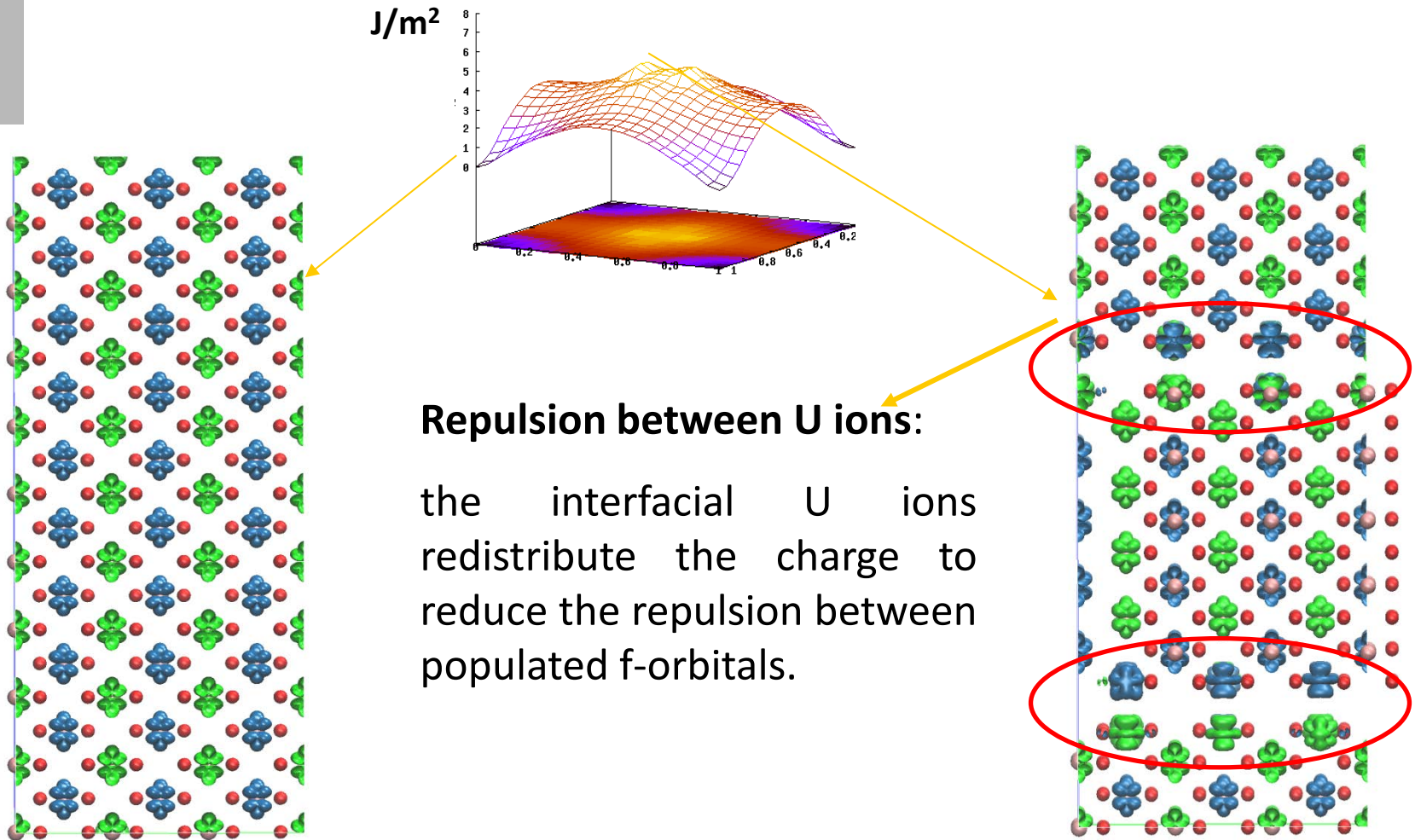
Force Field, Yakub



Fine grid of the 0.25-0.75 region
convergence w.r.t z axis

Why the DFT+U γ -surface is corrugated?

Electronic structure evolution during the {110} sampling

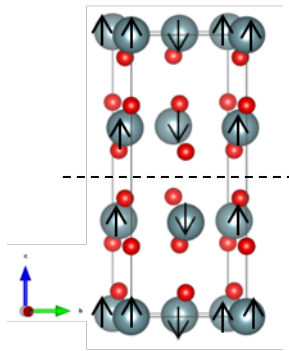


Validating Empirical Potentials with DFT+U

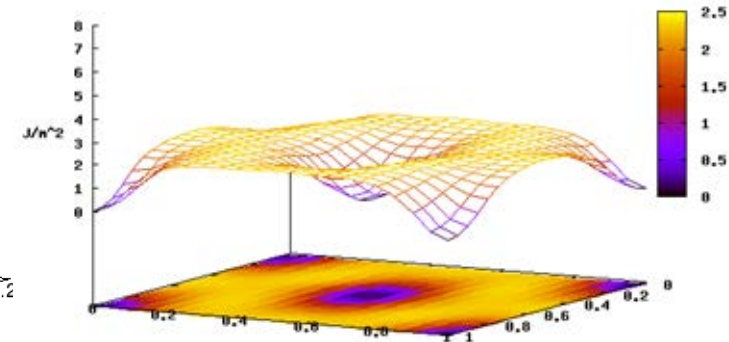
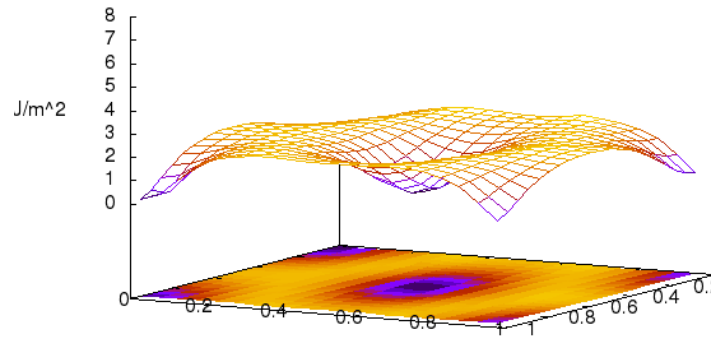
{111}

PBE+U

Force Field (Yakub)



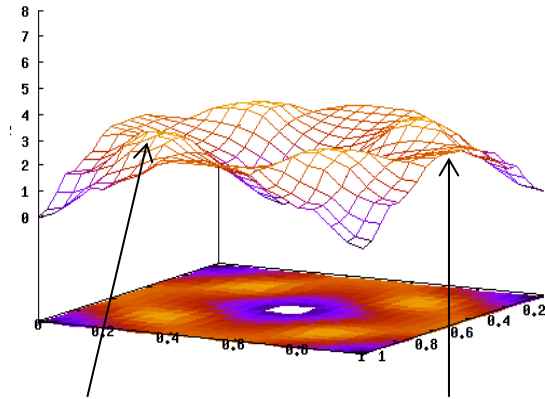
4x3x12 cell
2592 atoms (!)



- Qualitative and quantitative agreement between DFT+U and Empirical Potentials.
- The glide is governed by distortion of weak O-O interactions.

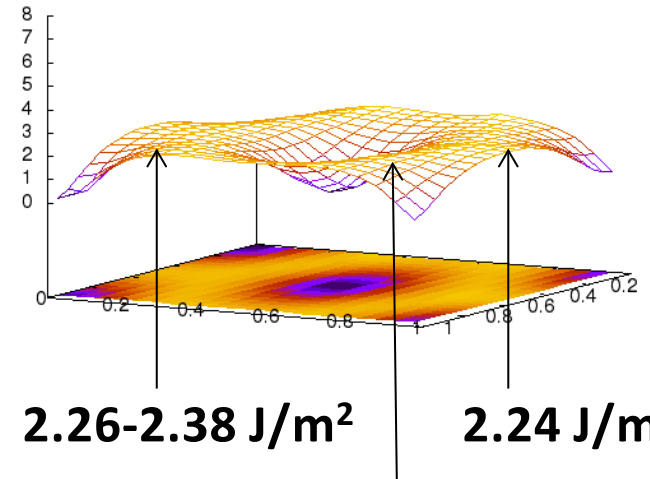
{001} and {111} lowest energy glides

{001}



2.75-3.09 J/m² 2.27-2.65 J/m²

{111}



2.26-2.38 J/m² 2.24 J/m²
2.23-2.24 J/m²

- {111} and {001} planes have comparable GSF energies, with {111} being slightly lower.
- The spread of DFT computed GSF energies: ~ 0.3 J/m², *f-orbital population scheme*.
- The spread of EP computed GSF energies: ~ 3 J/m².

Summary

- Set of tools for computing and analysing the γ – surfaces was devised.
- Capablity of modelling materials with complex electronic structure.
- DFT data: $\{001\}$ and $\{111\}$ γ -surfaces have comparable GSF energies, with $\{111\}$ having slightly lower values.



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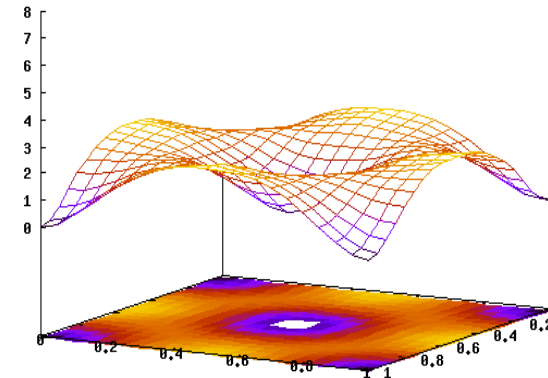
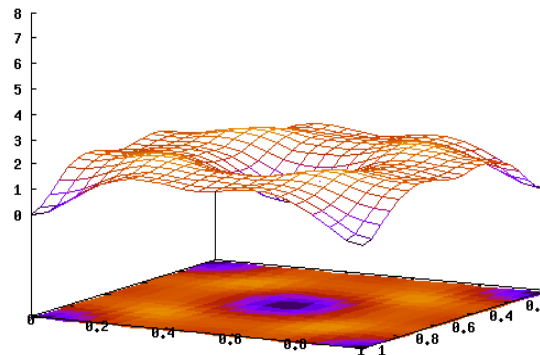
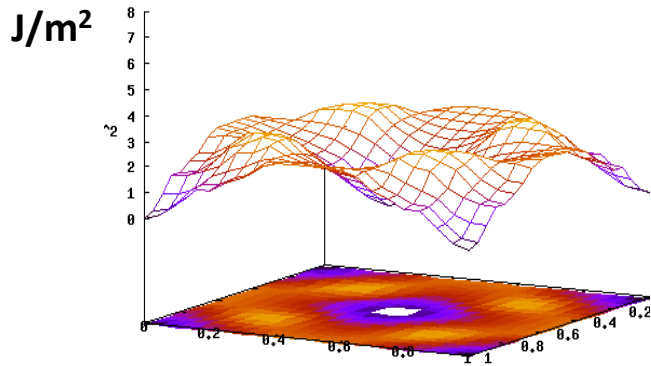


{001}

DFT+U

Force Field, Yakub

Force Field, Morelon



DFT+U results in a surface with shape similar to Yakub EP

Electronic degrees of freedom, *i.e.* the f-orbital populations, allow accommodation of lower energy geometric configurations.

Will be demonstrated for the {110} surface