

Studying disulfide damage at low X-ray doses with an engineered protein approach Timothy R. Stachowski ^{1,2,4,#} Mary E. Snell ¹ and Edward H. Snell ^{1,2,3}

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Abstract

A significant problem in biological X-ray crystallography is the radiation chemistry caused by the incident X-ray beam. This causes both global and site specific damage Global damage manifests itself in the decay of the diffraction pattern and data processing parameters. Site specific damage targets glutamates, aspartates, methionines, and disulfide bonds. This can misdirect the biological interpretation of the structural models produced. Cryo-cooling crystals has been successful in mitigating but not eliminating damage; however, cooling has also been shown to limit functionally relevant protein conformations. Due to this, and the difficulty in cryoprotecting some systems, there has been interest in the return to near-physiological temperature crystallography. Radiation chemistry at these conditions is less well studied. The doses used for X-ray crystallography under both cryocooled and near physiological temperature are in the kGy to MGy range. X-rays are used therapeutically at much lower doses. Disulfide bonds are among the most significantly affected species in a protein in the crystalline state but limited information is known about their response to damage in vivo. In this work we engineered a protein that dimerizes through a vulnerable disulfide bond to understand if radiation damage processes seen in structural studies translate to conditions closer to physiology, specifically in solution. We monitored monomerization with small angle solution X-ray scattering (SAXS), simultaneously pumping the sample with X-rays and while probing for structural impact using doses that are therapeutically relevant and a fraction of that required for crystallographic studies. Our results show that X-ray radiation drives a dose dependent fragmentation of the engineered protein that can be explained by a dimer to monomer transition, indicating disulfide bond cleavage. This supports the crystalline mechanism and suggests that crystallographic damage results can be extrapolated to physiologic conditions. Fragmentation was observed to be pH dependent, suggesting radiation damage processes and future routes for investigation and mitigation. The engineered protein approach represents a promising tool for advancing radiation damage studies and studying physiologically relevant radiation-protein interactions [1]. Irradiating the engineered protein in the presence of radicals provides insights into the underlying chemical mechanism [2].

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Radiation drives a change in the conformational ensemble of the dimer



shown in (d) for the distribution of D_{max} and (e) R_g . Dotted line represents the random pool. The color gradient corresponds to the magnitude of the absorbed dose (36.3 Gy-1.2 kGy) delivered across 33 x 0.3 sec exposures where vellow is low dose and black is high dose

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Question: Can we characterize X-ray induced disulfide bond cleavage with SAXS using low X-ray doses? Steps 1. Develop new data collection strategies to obtain reliable data from low (Gy vs. KGy and MGy) X-ray doses (Stachowski et al. JSR, 2019) 2 Develop a method to monitor a residue specific change with a low-resolution technique 3 Develop methods of quantifying and analyzing these data 4. Use the system to interrogate the underlying chemical mechanism Monitoring a dimer to monomer transition with SAXS Perturbation I(s) = $v_k I_k(s)$ SAXS is an ensemble technique and avg. MW: 60 kDa avg. MW: 30 kDa sensitive to changes in oligomeric introduce disulfide state However, SAXS is also a low resolution technique that cannot detect changes to specific residues. Breakage of a bond that causes a X-ravs x2 change in oligomeric state should be detectable with SAXS and report on changes to specific bonds

Objectives

Engineering a dimer linked by a susceptible disulfide









Radiation exposure drives protein fragmentation

Fig 2. X-ray solution scattering analysis indicating that fragmentation is radiation dose dependent. The color gradient corresponds to the magnitude of the absorbed does (36.3 G)-1.2 kGy) delivered across 33 x 0.3 sec exposures where yellow is low dose and black is high dose. EndoH:rvs was irradiated at pH 7.5 and the results shown are averages of two replicates at a concentration of 5.0 mg/ml. in (a) Juffer subtracted scattering curves whibit a dose-dependent decrease in intensity. In (b) residuels acatuated between the first exposure and each sequential exposure show that intensity extinuit a dose-dependent dorecase in initients); in (b) residuais calculated between the tirst exposure and each sequentia exposure show that intensity decreases at law-rad increases at high-p-with increasing absorbed dose. The arrow identifies an isoscatient point at q = 0.017 A-; characteristic of a transition between two states. In (c) a pairwise distance distribution plot, P(r), normalized to MW, of the first, last exposure, and difference between first and last exposure arows that the fraction of longer distances in the protein decreased. The stope of the Guiner region remains linear (d) but decreases, indicating that the size of the protein is decreasing. A Dimensionless Kratky plot (e) also indicates that the protein is changing shape during irradiation.

Radical scavengers change the magnitude of fragmentation



Fig 5. Scavengers exhibit concentration denenden changes in integrated intensity. The effectiveness of integrated each scavenger (ascorbic acid cysteine, and sodium nitrate) to inhibit fragmentation was judged comparing relative X-ray dose driven changes in the accumulated integrated intensity (All). This method provides the running total change ir integrated intensity as the tota X-ray dose increases As All decreases it corresponds to an increase in fragmentation. As All approaches zero it indicates a decrease in fragmentation. Each scavenger was tested at seven concentrations ranging from 50 nM - 50 mM and monitored from 22 Gy (yellow) to 726 Gy (purple). Each scavenger exhibited concentration effects Low concentrations (~50 nM - 5 µM) increased the amount of fragmentation while hi concentrations (~500 µM high mM) mitigated damage. The effectiveness of each scavenge at a particular concentration also differed.