9th Annual Ambient Pressure X-ray Photoelectron Spectroscopy Workhop



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In situ analysis of propane oxidation on Ru/CeO2 catalyst by NAP-XPS

Content

Supported ruthenium catalysts are considered promising for catalytic oxidation of volatile organic compounds (VOCs) [1], which can be dangerous for human health. A significant catalytic activity of Ru on cerium dioxide can be especially noticed in the oxidation of light hydrocarbons. Due to high oxygen storage capacity and redox properties, CeO2 is considered an "active" support [2, 3], which can, during oxidation processes, provide an abundance of active oxygen for redox reactions and thus increase the reaction rate.

Using near-ambient pressure X-ray photoelectron spectroscopy (NAP-XPS), the chemical state of ruthenium in Ru/CeO2 catalyst was studied under simulated realistic conditions of pre-treatments (calcination and reduction) and C3H8 oxidation, allowing better understanding of correlations between the chemical state of ruthenium and its activity in C3H8 oxidation. It is shown that the Ru/CeO2 interaction with an oxygen-rich atmosphere (C3H8+O2 (1:5)) at \geq 300 °C results in ruthenium oxidation to the volatile RuO4, leading to its homogeneous redispersion inside the powder catalyst and increased catalytic activity.

References:

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